

# CHEMISTRY MULTIPLE CHOICE QUESTIONS

## A. Atoms, molecules and stoichiometry

2002 -2019



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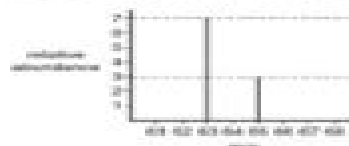
As a combustion, an alkane burns can be considered to have a fully set of 1.00 kg of methane (empirical formula  $\text{CH}_4$ ).

Which mass of carbon dioxide does a person produce each day if all the carbon they take in is digested and oxidised?

- A 0.007 kg    B 0.006 kg    C 1.30 kg    D 0.04 kg

(2002 NA/12)

The diagram shows the mass spectrum of a sample of naturally-occurring copper.



What is the relative atomic mass of this copper?

- A 63.2    B 63.5    C 63.6    D 64.0

(2002 NA/12)

Which pairs of compounds have the same empirical formula?

- ethane and ethene
- ethane and cyclohexane
- cyclohexane and oct-1-ene

(2002 NA/12)

A stream of  $10\text{cm}^3$  of methane and  $10\text{cm}^3$  of ethane was sparked with an excess of oxygen. After cooling to room temperature, the residual gas was passed through aqueous potassium hydroxide.

Which compounds have the empirical formula  $\text{CH}_2\text{O}$ ?

- methanol
- ethanoic acid
- methyl methanoate

(2002 ON/11)

What is the number of molecules in  $500\text{ cm}^3$  of oxygen under room conditions?

- A  $1.25 \times 10^{23}$   
 B  $1.25 \times 10^{24}$   
 C  $3.0 \times 10^{23}$   
 D  $3.0 \times 10^{24}$

(2004 MO/12)

In the preparation of soft margarine, glycerol trioleate



is fully hydrogenated so that, on average, one of its side-chains is converted into the  $\text{CH}_3(\text{CH}_2)_7\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CO}_2\text{C}_2\text{H}_5$  moiety and two side-chains are converted into the  $\text{CH}_3(\text{CH}_2)_7\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CO}_2\text{C}_2\text{H}_5$  moiety.

How many moles of hydrogen are required to convert one mole of glycerol trioleate into the soft margarine?

- A 4    B 5    C 6    D 8

(2008 MO/12)

Use of the Data Booklet is relevant to this question.

Analysed classes may obtain more useful accounts of extra work. Access to a 100% and

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**SA Adler**



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Benzene Chemical Compound MCQs covers topics of benzene arenes reaction phenol properties and reactions of phenol The chapter Carbonyl Compounds MCQs covers topics of carbonyl compounds aldehydes and ketone testing nucleophilic addition with HCN preparation of aldehydes and ketone reduction of aldehydes and ketone

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### **Table of Contents Chemistry Multiple Choice Questions With Answers**

1. Understanding the eBook Chemistry Multiple Choice Questions With Answers
  - The Rise of Digital Reading Chemistry Multiple Choice Questions With Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Multiple Choice Questions With Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemistry Multiple Choice Questions With Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Multiple Choice Questions With Answers

- Personalized Recommendations
  - Chemistry Multiple Choice Questions With Answers User Reviews and Ratings
  - Chemistry Multiple Choice Questions With Answers and Bestseller Lists
5. Accessing Chemistry Multiple Choice Questions With Answers Free and Paid eBooks
- Chemistry Multiple Choice Questions With Answers Public Domain eBooks
  - Chemistry Multiple Choice Questions With Answers eBook Subscription Services
  - Chemistry Multiple Choice Questions With Answers Budget-Friendly Options
6. Navigating Chemistry Multiple Choice Questions With Answers eBook Formats
- ePub, PDF, MOBI, and More
  - Chemistry Multiple Choice Questions With Answers Compatibility with Devices
  - Chemistry Multiple Choice Questions With Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Chemistry Multiple Choice Questions With Answers
  - Highlighting and Note-Taking Chemistry Multiple Choice Questions With Answers
  - Interactive Elements Chemistry Multiple Choice Questions With Answers
8. Staying Engaged with Chemistry Multiple Choice Questions With Answers
- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chemistry Multiple Choice Questions With Answers
9. Balancing eBooks and Physical Books Chemistry Multiple Choice Questions With Answers
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chemistry Multiple Choice Questions With Answers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Multiple Choice Questions With Answers
- Setting Reading Goals Chemistry Multiple Choice Questions With Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Multiple Choice Questions With Answers

- Fact-Checking eBook Content of Chemistry Multiple Choice Questions With Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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