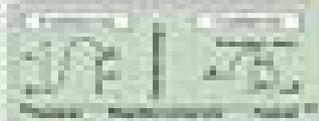


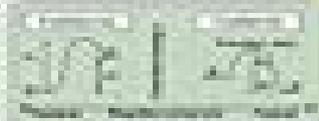
Equilibrium

- **Equilibrium** is a state where the net force on an object is zero.
 - An object can be in equilibrium even if it is moving.
 - An object can be in equilibrium even if it is accelerating.
- **Newton's First Law of Motion** states that an object at rest stays at rest and an object in motion stays in motion with the same speed and in the same direction unless acted upon by an unbalanced force.
 - An object at rest stays at rest.
 - An object in motion stays in motion with the same speed and in the same direction.
- **Newton's Second Law of Motion** states that the acceleration of an object is directly proportional to the net force acting on it and is inversely proportional to its mass.
 - $F = ma$
 - $a = \frac{F}{m}$
 - $F = \frac{ma}{m}$
- **Newton's Third Law of Motion** states that for every action, there is an equal and opposite reaction.
 - If object A exerts a force on object B, then object B exerts a force of equal magnitude and opposite direction on object A.



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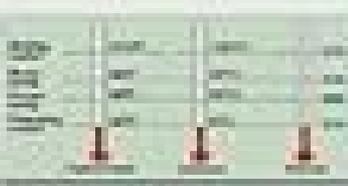
Nuclear Chemistry

- **Nuclear Chemistry** is the study of the changes in the nucleus of an atom.
 - It involves the emission of particles and energy from the nucleus.
 - It includes the study of radioactive decay and nuclear reactions.
- **Types of Processes**
 - **Radioactive Decay**: The spontaneous emission of particles and energy from the nucleus.
 - **Nuclear Fission**: The splitting of a heavy nucleus into two lighter nuclei, releasing energy.
 - **Nuclear Fusion**: The combining of two light nuclei to form a heavier nucleus, releasing energy.
- **Radioactive Decay**
 - **Alpha Decay**: The emission of an alpha particle (^4_2He).
 - **Beta Decay**: The emission of a beta particle ($^0_{-1}\text{e}$).
 - **Gamma Decay**: The emission of a gamma ray (γ).




Measurement & Units

- **Measurement & Units** is the process of determining the magnitude of a physical quantity.
 - It involves the use of a standard unit.
 - It includes the study of the SI system and other units.
- **SI Units**
 - **Length**: meter (m)
 - **Mass**: kilogram (kg)
 - **Time**: second (s)
 - **Temperature**: kelvin (K)
 - **Amount of Substance**: mole (mol)
 - **Electric Current**: ampere (A)
 - **Luminous Intensity**: candela (cd)
- **Conversion**
 - $1 \text{ km} = 1000 \text{ m}$
 - $1 \text{ kg} = 1000 \text{ g}$
 - $1 \text{ s} = 60 \text{ min} = 3600 \text{ s}$




Chemistry Quick Study Academic

Jin-Ying Zhang



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