

## Balancing & Reactions Test SCH3U

Name: \_\_\_\_\_

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**10**    **20**    **5**    **4**

Multiple Choice: Circle the most correct answer. (10K)

1. Which equation is improperly balanced?

- A)  $2 \text{KNO}_3 \rightarrow 2 \text{KNO}_2 + \text{O}_2$                       B)  $\text{H}_2\text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_2$   
C)  $2 \text{Na}_2\text{O}_2 + 2 \text{H}_2\text{O} \rightarrow 4 \text{NaOH} + \text{O}_2$     D)  $2\text{H}_2\text{O} \rightarrow 2\text{H}_2 + \text{O}_2$

2. In every balanced chemical equation, each side of the equation has the same number of \_\_\_\_\_.

- A) atoms of each element    B) molecules                      C) Moles                      D) coefficients

3. Which metal would ggg displace lead, Pb, in a lead nitrate,  $\text{Pb}(\text{NO}_3)_2$ , solution?

- A) copper, Cu    B) magnesium, Mg    C) potassium, K                      D) lithium, Li    E) none of the above

4. Which of the following are decomposition reactions?



- A) 1, 2, and 3                      B) 4 only                      C) 1, 2, 3, and 4                      D) 2 and 3                      E) 2, 3, and 4

5. Predict the product in the synthesis reaction of Al with  $\text{N}_2$ .

- A) AlN    B)  $\text{Al}_3\text{N}$     C)  $\text{AlN}_2$     D)  $\text{Al}_2\text{N}_3$     E)  $\text{AlN}_3$

6. What is true of the oxide formed when magnesium reacts with oxygen?

- A) Its formula is  $\text{Mg}_2\text{O}$                       C) It is molecular  
B) It is basic                                      D) It is amphoteric

7. Oxides of the active metals combine with acid to form \_\_\_\_\_.

- A) H<sub>2</sub> gas    B) metal hydrides    C) water and a salt    D) oxygen gas    E) metal hydroxides

8. Which one of the following compounds would produce an acidic solution when dissolved in water?

- A)  $\text{Na}_2\text{O}$     B) CaO    C) MgO    D)  $\text{CO}_2$     E) SeO

9. In a double-displacement reaction, the \_\_\_\_\_.

- A) products are always molecular    B) reactants are usually two ionic compounds  
C) reactants are two elements    D) products are a new element and a new compound

# Chemistry Chapter 11 Chemical Reactions Test Answers

**M Mark**



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