

Chapter Test II

A. Matching

Match each item in Column I with the correct description in Column II. Write the letter of the correct description.

Column I

- _____ 1. An oxidation-reduction reaction in which an element is oxidized and another element is reduced.
- _____ 2. The number of protons in the nucleus of an atom of an element.
- _____ 3. The number of protons in an element's nucleus.
- _____ 4. The ratio of the actual weight of an element to the standard atomic weight.
- _____ 5. The number of protons and neutrons in the nucleus of an atom of an element.
- _____ 6. The number of protons and neutrons in the nucleus of an atom of an element.

Column II

- a. atomic weight
- b. limiting reagent
- c. molecular weight
- d. molar mass
- e. percent yield
- f. atomic number
- g. percent composition

B. Multiple Choice

Choose the best answer for each question.

- _____ 1. The number of protons in an atom of an element is the same as the number of
 - a. neutrons in the nucleus
 - b. electrons in the atom
 - c. total mass in the nucleus
 - d. total mass in the atom
- _____ 2. Which of the following is not a chemical reaction?
 - a. $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
 - b. $\text{H}_2\text{O} \rightarrow \text{H}_2 + \text{O}_2$
 - c. $\text{H}_2\text{O} + \text{CO}_2 \rightarrow \text{H}_2\text{CO}_3$
 - d. $\text{H}_2\text{O} + \text{H}_2\text{O} \rightarrow \text{H}_2\text{O}$
- _____ 3. The number of protons in the nucleus of an atom of an element is the same as the number of
 - a. neutrons in the nucleus
 - b. electrons in the atom
 - c. total mass in the nucleus
 - d. total mass in the atom

Chemistry Chapter 12 Test B Stoichiometry

S Ashworth

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