

Unit 4 - Chemical Bonding Practice Test

Name: _____ Class: _____ Date: _____

Part I. Multiple choice

- The atoms of which element will form cations?
A. selenium B. neon C. copper D. iodine
- The atoms of which element will gain electrons to form an ion?
A. mercury B. bromine C. beryllium D. potassium
- An atom of calcium becomes a cation when it forms an ionic bond with two atoms of chlorine. What is the charge on the calcium ion?
A. 1+ B. 2+ C. 3+ D. 4+
- Which combination of elements will form an ionic bond?
A. oxygen and fluorine B. nitrogen and helium
C. sodium and calcium D. potassium and chlorine
- Predict the formula of lithium nitride.
A. LiN B. Li₃N C. Li₂N D. LiN₃
- What is the correct name of this compound, Pb(CO₃)₂?
A. Lead Carbonate B. Lead Carbonite
C. Lead (II) Carbonite D. Lead (IV) Carbonate
- How many pairs of electrons are shared in a double bond?
A. 1 B. 2 C. 3 D. 4
- Covalent bonds are most likely to form between atoms of which types of elements?
A. metals and metalloids B. metals and metals
C. metals and non-metals D. non-metals and non-metals
- What happens to electrons during covalent bonding?
A. They are given up by atoms. B. They are shared between atoms.
C. They are transferred between atoms. D. They are taken in by atoms.

Chemistry Chemical Bonding Test Answers

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