

# CHEMISTRY VIVA QUESTIONS WITH ANSWERS CLASS XII

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## [1] VOLUMETRIC ANALYSIS

**Question.1.What is a standard solution?**

**Answer.** A solution whose strength is known is called a standard solution.

**Question.2.What is a normal solution?**

**Answer.** A solution containing one gram-equivalent mass of the solute per litre of the solution is called a normal solution.

**Question.3.What is the equivalent mass of  $\text{KMnO}_4$  when it acts as oxidizing agent in acidic medium?**

**Answer.**  $\text{KMnO}_4$  loses 5 electrons per molecule, when it acts as oxidizing agent in the presence of acids. Therefore, its equivalent mass is one-fifth of its molecular mass.  
volumetric-analysis-viva-questions-with-answers-1

**Question.4.Is sodium hydroxide a primary standard?**

**Answer.** No.

**Question.5.Are 'molality' and 'molarity' same?**

**Answer.** No, molality of a solution is defined as the number of moles of solute present in 1000 grams of the solution whereas molarity tells us about the number of moles of the solute present per litre of the solution.

**Question.6.What would be the normality of 0.10M  $\text{KMnO}_4$ ?**

**Answer.** It will be  $0.1 \times 5 = 0.5 \text{ N}$ .

**Question.7.What volume of 10M HCl must be diluted with water to get 1L of 1M HCl?**

**Answer.** 0.1L.

**Question.8.What is the basicity of  $\text{H}_2\text{SO}_4$ ?**

**Answer.** 2.

**Question.9.What is the relationship between normality (N), molarity (M), molecular mass and equivalent mass?**

**Answer.** Normality  $\times$  Eq. Mass = Molarity  $\times$  Mol. Mass.

**Question.10.Why front door of the balance is closed during weighing?**

**Answer.** Opening the front door causes vibrations in the pan due to operator's breath which leads to inaccurate results.

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### **Table of Contents Chemistry Practical Viva Questions With Answers**

1. Understanding the eBook Chemistry Practical Viva Questions With Answers
  - The Rise of Digital Reading Chemistry Practical Viva Questions With Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Practical Viva Questions With Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemistry Practical Viva Questions With Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Practical Viva Questions With Answers
  - Personalized Recommendations
  - Chemistry Practical Viva Questions With Answers User Reviews and Ratings

- Chemistry Practical Viva Questions With Answers and Bestseller Lists
- 5. Accessing Chemistry Practical Viva Questions With Answers Free and Paid eBooks
  - Chemistry Practical Viva Questions With Answers Public Domain eBooks
  - Chemistry Practical Viva Questions With Answers eBook Subscription Services
  - Chemistry Practical Viva Questions With Answers Budget-Friendly Options
- 6. Navigating Chemistry Practical Viva Questions With Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Chemistry Practical Viva Questions With Answers Compatibility with Devices
  - Chemistry Practical Viva Questions With Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chemistry Practical Viva Questions With Answers
  - Highlighting and Note-Taking Chemistry Practical Viva Questions With Answers
  - Interactive Elements Chemistry Practical Viva Questions With Answers
- 8. Staying Engaged with Chemistry Practical Viva Questions With Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chemistry Practical Viva Questions With Answers
- 9. Balancing eBooks and Physical Books Chemistry Practical Viva Questions With Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chemistry Practical Viva Questions With Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Practical Viva Questions With Answers
  - Setting Reading Goals Chemistry Practical Viva Questions With Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Practical Viva Questions With Answers
  - Fact-Checking eBook Content of Chemistry Practical Viva Questions With Answers
  - Distinguishing Credible Sources

13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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