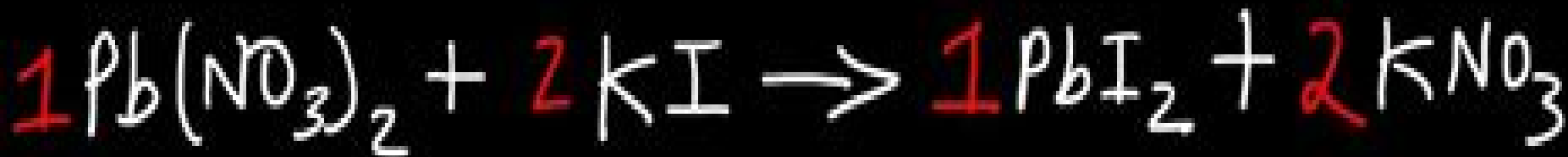


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$$\frac{0.29 \text{ mol KI}}{\cancel{2} \times} \times \frac{0.037 \text{ L}}{1} \times \frac{1 \text{ mol Pb}(\text{NO}_3)_2}{\cancel{2} \text{ mol KI}} \times \frac{1}{0.042 \text{ L}}$$

$$= 0.1277 \text{ M}$$

$$M = \frac{\text{mol}}{\text{L}}$$

Chemistry Solution Stoichiometry

J Ma



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