

Introduction 11

An Electron's Address

Name: _____
 Date: _____
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Introduction: Energy Levels and Sublevels

In previous lessons you have learned about the Bohr model of the atom. The Bohr model of the atom says that electrons orbit the nucleus in fixed energy levels. The Bohr model of the atom says that electrons orbit the nucleus in fixed energy levels. Bohr's model of the atom says that electrons orbit the nucleus in fixed energy levels.

Bohr's model of the atom says that electrons orbit the nucleus in fixed energy levels. Bohr's model of the atom says that electrons orbit the nucleus in fixed energy levels. Bohr's model of the atom says that electrons orbit the nucleus in fixed energy levels.

Table 1

Energy Level	Sublevels in each energy level
1	s
2	s, p
3	s, p, d
4	s, p, d, f

From the data in Table 1, you can see that the number of sublevels in each energy level increases as the energy level increases. The number of sublevels in each energy level increases as the energy level increases.

Critical Thinking Questions

- How many sublevels are in the 1st energy level?
 The sublevels in the 1st energy level are the s sublevel. The s sublevel is the only sublevel in the 1st energy level.
- How many sublevels are in the 2nd energy level?
 The sublevels in the 2nd energy level are the s and p sublevels. The s and p sublevels are the only sublevels in the 2nd energy level.
- How many sublevels are in the 3rd energy level?
 The sublevels in the 3rd energy level are the s, p, and d sublevels. The s, p, and d sublevels are the only sublevels in the 3rd energy level.
- How many sublevels are in the 4th energy level?
 The sublevels in the 4th energy level are the s, p, d, and f sublevels. The s, p, d, and f sublevels are the only sublevels in the 4th energy level.
- How many sublevels are in the 5th energy level?
 The sublevels in the 5th energy level are the s, p, d, and f sublevels. The s, p, d, and f sublevels are the only sublevels in the 5th energy level.

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Ensheng Dong



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