

Richard I. Henderson



# Compensation Management in a Knowledge-Based World

Tenth Edition

# Compensation Management In A Knowledge Based World 10th Tenth Edition

**IM Harris**



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web during oxidation and reduction the oxidation numbers of atoms change if an atom is oxidized its oxidation number increases ie it becomes more ve or less ve if an atom is reduced its oxidation number decreases ie it becomes less ve or more ve

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web write the oxidation and reduction reactions for each redox reaction the first one is done for you  $23 \text{ Fe} + \text{CO}_2 \rightarrow \text{Fe}_2\text{O}_3 + \text{C}$   
oxidation  $\text{CO} + \text{CO}_2 + 2\text{e}^-$  reduction  $\text{Fe} + 2\text{e}^- \rightarrow \text{Fe}^{2+}$   $24 \text{ 3 Ag} + \text{Ni}^{3+} \rightarrow 3 \text{ Ag} + \text{Ni}$  oxidation reduction  $25 \text{ Cu} + \text{Pb}^{2+} \rightarrow \text{Cu}^{2+} + \text{Pb}$  oxidation reduction  $26 \text{ O}_2$

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 $\text{b} \text{ 5CO} + \text{I}_2\text{O}_5 \rightarrow 5\text{CO}_2 + \text{I}_2$   
 $\text{c} \text{ O}_2 + \text{I}_2\text{O}_5 \rightarrow \text{CO}_2$

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