

Date 18-MAR-2023 Name Kayla Vines
 Section 2230-1679 Team _____
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REPORT SHEET | LAB
**Compounds and
 Their Bonds** | **9**

A. Ions: Transfer of Electrons

| Element | Atomic Number | 1. Electron Configuration of Atom | 2. Electron-Dot Symbol | 3. Loss or Gain of Electrons | 4. Electron Configuration of Ion | 5. Ionic Charge | 6. Symbol of Ion | 7. Name of Ion |
|----------|---------------|-----------------------------------|------------------------|------------------------------|----------------------------------|-----------------|------------------|----------------|
| Sodium | 11 | $1s^2 2s^2 2p^6 3s^1$ | Na [•] | lose 1 e ⁻ | $1s^2 2s^2 2p^6$ | 1+ | Na ⁺ | Sodium |
| Nitrogen | 7 | $1s^2 2s^2 2p^3$ | $:\ddot{N}:$ | gains 3e ⁻ | $1s^2 2s^2 2p^6$ | 3- | N ³⁻ | nitride |
| Aluminum | 13 | $1s^2 2s^2 2p^6 3s^2 3p^1$ | $\cdot\text{Al}\cdot$ | lose 3e ⁻ | $1s^2 2s^2 2p^6$ | 3+ | Al ³⁺ | aluminum |
| Chlorine | 17 | $1s^2 2s^2 2p^6 3s^2 3p^5$ | $:\ddot{Cl}\cdot$ | gains 1e ⁻ | $1s^2 2s^2 2p^6 3s^2 3p^6$ | 1- | Cl ⁻ | chloride |
| Calcium | 20 | $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2$ | Ca ^{••} | lose 2e ⁻ | $1s^2 2s^2 2p^6 3s^2 3p^6$ | 2+ | Ca ²⁺ | calcium |
| Oxygen | 8 | $1s^2 2s^2 2p^4$ | $:\ddot{O}:$ | gains 2e ⁻ | $1s^2 2s^2 2p^6$ | 2- | O ²⁻ | oxide |

* $1s^2 2s^2 2p^6 3s^2 3p^6$

B. Ionic Compounds and Formulas

1. Physical properties

| Compound | Appearance | Density | Melting Point |
|-----------------------|-------------------------|------------------------|---------------|
| Sodium chloride, NaCl | white crystalline solid | 2.17 g/cm ³ | 1,474 °F |

2. Formulas of ionic compounds

| Name | Positive Ion | Negative Ion | Formula |
|-------------------|------------------|-----------------|--------------------------------|
| Lithium iodide | Li ⁺ | I ⁻ | LiI |
| Aluminum oxide | Al ³⁺ | O ²⁻ | Al ₂ O ₃ |
| Calcium sulfide | Ca ²⁺ | S ²⁻ | CaS |
| Magnesium bromide | Mg ²⁺ | Br ⁻ | MgBr ₂ |
| Potassium nitride | K ⁺ | N ³⁻ | K ₃ N |
| Sodium fluoride | Na ⁺ | F ⁻ | NaF |

Compounds And Their Formulas Report Sheet Answers

Katrin Zwirglmaier



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