

## Covalent Bonding Review Sheet – Dec 12, 2012

Name \_\_\_\_\_ class period \_\_\_\_\_

- Given the formula, a compound can be named using conventional systems that include Greek prefixes and Roman numeral where appropriate.
- Differences in electronegativity values can be used to predict where a bond fits on the continuum between ionic and covalent bonds.
- Since most compounds contain multiple bonds, a substance may contain more than one type of bond.

### Section 1: Naming and Formula Writing

Write the formulas for the following covalent compounds

1. Nitrogen Monoxide NO
2. Dinitrogen Monoxide N<sub>2</sub>O
3. Nitrogen Dioxide NO<sub>2</sub>
4. Dinitrogen Trioxide N<sub>2</sub>O<sub>3</sub>
5. Dinitrogen Tetraoxide N<sub>2</sub>O<sub>4</sub>
6. Dinitrogen Pentaoxide N<sub>2</sub>O<sub>5</sub>
7. carbon dioxide CO<sub>2</sub>
8. phosphorus triiodide PI<sub>3</sub>
9. sulfur dichloride SCl<sub>2</sub>
10. nitrogen trifluoride NF<sub>3</sub>

Write names from the following formulas

11. N<sub>2</sub>F<sub>4</sub> dinitrogen tetrafluoride
12. SCl<sub>4</sub> sulfur tetrachloride
13. ClF<sub>3</sub> chlorine trifluoride
14. SiO<sub>2</sub> silicon dioxide
15. P<sub>4</sub>O<sub>10</sub> tetraphosphorus decaoxide
16. SiF<sub>4</sub> silicon tetrafluoride
17. H<sub>2</sub>O dihydrogen monoxide
18. Br<sub>2</sub> dibromide or bromine
19. HBr hydrogen monobromide
20. N<sub>2</sub>S<sub>3</sub> dinitrogen trisulfide

# Concept Review Section Covalent Bonds Answer Key

**Kenneth R. Miller**



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**Prentice Hall Science Explorer** Michael J. Padilla, 2002 Modules McDougal Littell Incorporated, 2005 *Science Books & Films* , 1982 **Biochemistry Review** David M. Glick, 1975 Previous editions by E H Mosbach **Biology** Neil A. Campbell, Jane B. Reece, 2005 Neil Campbell and Jane Reece s BIOLOGY remains unsurpassed as the most successful majors biology textbook in the world This text has invited more than 4 million students into the study of this dynamic and essential discipline The authors have restructured each chapter around a conceptual framework of five or six big ideas An Overview draws students in and sets the stage for the rest of the chapter each numbered Concept Head announces the beginning of a new concept and Concept Check questions at the end of each chapter encourage students to assess their mastery of a given concept New Inquiry Figures focus students on the experimental process and new Research Method Figures illustrate important techniques in biology Each chapter ends with a Scientific Inquiry Question that asks students to apply scientific

investigation skills to the content of the chapter      *Chemical principles* Paul B. Kelter, 1998

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