

Name : _____ Date : _____

Lab Periodic Trends Worksheet

The atomic and the ionic radius of the first 18 elements of the periodic table are given below.

Element	H	He	Li	Be	B	C	N	O	F
Atomic Number	1	2	3	4	5	6	7	8	9
Atomic Radius (Å)	0.79	0.49	2.05	1.4	1.17	0.91	0.75	0.65	0.57
Ionic Radius (Å)	0.012	-	0.76	0.35	0.23	0.16	0.13	1.4	1.33

Ne	Na	Mg	Al	Si	P	S	Cl	Ar
10	11	12	13	14	15	16	17	18
0.38	2.23	1.72	1.82	1.46	1.23	1.09	0.97	0.71
-	1.02	0.72	0.535	0.4	0.38	0.37	1.81	-

Using the data above, construct the line graph of atomic radius vs. atomic number and ionic radius vs. the atomic number.

Atomic Radius vs. Atomic Number



Ionic Radius vs. Atomic Number



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**Robert C. Smoot, Smoot, Richard G.
Smith, Jack Price**

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Exploring General Chemistry in the Laboratory Colleen F. Craig, Kim N. Gunnerson, 2017-02-01 This laboratory manual is intended for a two semester general chemistry course. The procedures are written with the goal of simplifying a complicated and often challenging subject for students by applying concepts to everyday life. This lab manual covers topics such as composition of compounds, reactivity, stoichiometry, limiting reactants, gas laws, calorimetry, periodic trends, molecular structure, spectroscopy, kinetics, equilibria, thermodynamics, electrochemistry, intermolecular forces, solutions, and coordination complexes. By the end of this course, you should have a solid understanding of the basic concepts of chemistry, which will give you confidence as you embark on your career in science.

O-level Chemistry Challenging Drill Questions (Concise) (Yellowreef) Thomas Bond, Chris Hughes, 2013-11-07 questions from top schools since 2003 complete encyclopedia of over 350 questions complete answer keys full set of step by step solution approaches available separately complete edition and concise edition eBooks available

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