

**STUDENT BOOK ANSWERS**
**To come**
**Chapter 9: Gas laws**
**Worked example 9.1**

- 1  $n = \frac{V}{V_m} = \frac{2.33}{22.71} = 0.1026 \text{ mol}$   
 $m = n \times MM = 0.1026 \times (12.01 + (4 \times 1.008)) = 1.646 = 1.65 \text{ g (3sf)}$
- 2  $n = \frac{m}{MM} = \frac{23.0}{14.01 + (3 \times 1.008)} = 1.350 \text{ mol}$   
 $V = n \times V_m = 1.350 \times 24.79 = 33.47 = 33.5 \text{ L (3sf)}$

**Worked example 9.2**

- 1  $2\text{Na(s)} + 2\text{H}_2\text{O(l)} \rightarrow 2\text{NaOH(aq)} + \text{H}_2\text{(g)}$   
 $n(\text{Na}) = \frac{m}{MM} = \frac{1.87}{22.99 + 16.00 + 1.008} = 0.04675 \text{ mol}$   
 $n(\text{H}_2) = \frac{1}{2} \times n(\text{Na}) = 0.5 \times 0.04675 = 0.02338 \text{ mol}$   
 $V = n \times V_m = 0.02338 \times 24.79 = 0.5795 = 0.580 \text{ L (3sf)}$
- 2  $n(\text{HCl}) = \frac{V}{V_m} = \frac{13.4}{22.71} = 0.5900 \text{ mol}$   
 $2\text{KCl(aq)} + \text{H}_2\text{SO}_4\text{(aq)} \rightarrow 2\text{HCl(g)} + \text{K}_2\text{SO}_4\text{(aq)}$   
 $n(\text{KCl}) = n(\text{HCl}) = 0.5900 \text{ mol}$   
 $m(\text{KCl}) = n \times MM = 0.5900 \times (39.10 + 35.45) = 43.98 = 44.0 \text{ g (3sf)}$

**Check your understanding 9.1/9.2**

- 1 The volume of a gas changes with temperature and pressure, both of which compress or expand the gas particles.
- 2 Use the formula  $n = \frac{V}{V_m}$  to convert the volume of gas to a number of moles.  $V_m$  depends upon the temperature of the gas. Use  $m = n \times MM$  to convert the number of moles ( $n$ ) to a mass ( $m$ ) by multiplying by the molecular mass ( $MM$ ).
- 3 **a** **i**  $2 \times 250 = 500 \text{ mL}$   
**ii**  $250 \text{ mL}$   
**b** **i**  $500 \text{ mL}$   
**ii** Nothing, as the law of combining volumes only applies to gases.

# Chapter 9 Notes Cengage Learning

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## Chapter 9 Notes Cengage Learning:

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maintain timely coverage of current marketing trends and strategies      Executive MBA in IT - City of London College of Economics - 12 months - 100% online / self-paced City of London College of Economics, Overview An MBA in information technology or a Master of Business Administration in Information Technology is a degree that will prepare you to be a leader in the IT industry Content Managing Projects and IT Information Systems and Information Technology IT Manager s Handbook Business Process Management Human Resource Management Principles of Marketing The Leadership Just What Does an IT Manager Do The Strategic Value of the IT Department Developing an IT Strategy Starting Your New Job The First 100 Days etc Managing Operations Cut Over into Operations Agile Scrum Project Management IT Portfolio Management The IT Organization etc Introduction to Project Management The Project Management and Information Technology Context The Project Management Process Groups A Case Study Project Integration Management Project Scope Management Project Time Management Project Cost Management Project Quality Management Project Human Resource Management Project Communications Management Project Risk Management Project Procurement Management Project Stakeholder Management 50 Models for Strategic Thinking English Vocabulary For Computers and Information Technology Duration 12 months Assessment The assessment will take place on the basis of one assignment at the end of the course Tell us when you feel ready to take the exam and we ll send you the assignment questions Study material The study material will be provided in separate files by email download link      **Student Learning Guide to Accompany Effective Marketing** William G. Zikmund,2001-02 This comprehensive learning guide reinforces the major topics in each chapter through the use of the following study tools chapter summaries vocabulary building exercises true false and multiple choice questions and critical thinking activities In addition crossword puzzles using key terms from each chapter help reinforce the definitions of marketing terminology      *Understanding Health Insurance* Brisky,Ruth M. Burke,Michelle A. Green,Jo Ann C. Rowell,Lynette M. Williamson,Jo Ann C Rowell,2007-12      **Education** Cengage Gale,2010-04-09 The Information Plus Reference Series provides statistical data on 32 of today s most controversial and most studied social issues Each Information Plus title is a compilation of current and historical statistics with analysis on aspects of one contemporary social issue such as abortion capital punishment and genetic engineering Each title is divided into chapters that are devoted to a particular topic The text provides a clear and comprehensive summary of up to date research on the topic and is interspersed with the statistical tables charts and graphs Each table is directly referred to and carefully explained in the text

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### **Table of Contents Chapter 9 Notes Cengage Learning**

1. Understanding the eBook Chapter 9 Notes Cengage Learning
  - The Rise of Digital Reading Chapter 9 Notes Cengage Learning
  - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 9 Notes Cengage Learning
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chapter 9 Notes Cengage Learning
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 9 Notes Cengage Learning
  - Personalized Recommendations
  - Chapter 9 Notes Cengage Learning User Reviews and Ratings

- Chapter 9 Notes Cengage Learning and Bestseller Lists
- 5. Accessing Chapter 9 Notes Cengage Learning Free and Paid eBooks
  - Chapter 9 Notes Cengage Learning Public Domain eBooks
  - Chapter 9 Notes Cengage Learning eBook Subscription Services
  - Chapter 9 Notes Cengage Learning Budget-Friendly Options
- 6. Navigating Chapter 9 Notes Cengage Learning eBook Formats
  - ePub, PDF, MOBI, and More
  - Chapter 9 Notes Cengage Learning Compatibility with Devices
  - Chapter 9 Notes Cengage Learning Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chapter 9 Notes Cengage Learning
  - Highlighting and Note-Taking Chapter 9 Notes Cengage Learning
  - Interactive Elements Chapter 9 Notes Cengage Learning
- 8. Staying Engaged with Chapter 9 Notes Cengage Learning
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chapter 9 Notes Cengage Learning
- 9. Balancing eBooks and Physical Books Chapter 9 Notes Cengage Learning
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chapter 9 Notes Cengage Learning
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Chapter 9 Notes Cengage Learning
  - Setting Reading Goals Chapter 9 Notes Cengage Learning
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chapter 9 Notes Cengage Learning
  - Fact-Checking eBook Content of Chapter 9 Notes Cengage Learning
  - Distinguishing Credible Sources

13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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