

**UNIT 10: PERIODIC TABLE** **STANDARD CURRICULAR PERFORMANCE INDICATOR**

**Section 1.1 (continued)**

In your notebook, read aloud the modern periodic table.

Copy the letterheads in the boxes on the left below from the periodic table to complete the table on the right.

_____
_____
_____
_____
_____

Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, under the letter of the matching item in Column B.

**Column A**

**Column B**

- |   |                                  |
|---|----------------------------------|
| _____ 16. A column in the periodic table                  | _____ a. metals                  |
| _____ 17. A row in the periodic table                     | _____ b. group                   |
| _____ 18. Group 1 elements                                | _____ c. period                  |
| _____ 19. Elements that are shiny and conduct electricity | _____ d. representative elements |
| _____ 20. Group 8 elements                                | _____ e. transition elements     |

In the space on the left, write each of the statements in terms of the statement in bold. Change the underlined word or phrase to make it true.

- \_\_\_\_\_ 21. There are two representative elements of platinum.
- \_\_\_\_\_ 22. Most elements besides boron of the elements in the periodic table are metalloids.
- \_\_\_\_\_ 23. Group 1A elements except for hydrogen are located on the left side.
- \_\_\_\_\_ 24. Group 2A elements are the alkaline earth metals.
- \_\_\_\_\_ 25. Group 7A elements are highly reactive nonmetals known as halogens.
- \_\_\_\_\_ 26. Group 8A elements are (a) unreactive elements known as noble gases.
- \_\_\_\_\_ 27. Metalloids have properties of both metals and nonmetals.

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**David Kirk**



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## **Table of Contents Chapter 5 Study Guide For Content Mastery Answer Key Chemistry**

1. Understanding the eBook Chapter 5 Study Guide For Content Mastery Answer Key Chemistry
  - The Rise of Digital Reading Chapter 5 Study Guide For Content Mastery Answer Key Chemistry
  - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 5 Study Guide For Content Mastery Answer Key Chemistry
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chapter 5 Study Guide For Content Mastery Answer Key Chemistry
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 5 Study Guide For Content Mastery Answer Key Chemistry
  - Personalized Recommendations
  - Chapter 5 Study Guide For Content Mastery Answer Key Chemistry User Reviews and Ratings
  - Chapter 5 Study Guide For Content Mastery Answer Key Chemistry and Bestseller Lists
5. Accessing Chapter 5 Study Guide For Content Mastery Answer Key Chemistry Free and Paid eBooks
  - Chapter 5 Study Guide For Content Mastery Answer Key Chemistry Public Domain eBooks
  - Chapter 5 Study Guide For Content Mastery Answer Key Chemistry eBook Subscription Services
  - Chapter 5 Study Guide For Content Mastery Answer Key Chemistry Budget-Friendly Options
6. Navigating Chapter 5 Study Guide For Content Mastery Answer Key Chemistry eBook Formats
  - ePub, PDF, MOBI, and More
  - Chapter 5 Study Guide For Content Mastery Answer Key Chemistry Compatibility with Devices
  - Chapter 5 Study Guide For Content Mastery Answer Key Chemistry Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chapter 5 Study Guide For Content Mastery Answer Key Chemistry
  - Highlighting and Note-Taking Chapter 5 Study Guide For Content Mastery Answer Key Chemistry
  - Interactive Elements Chapter 5 Study Guide For Content Mastery Answer Key Chemistry

8. Staying Engaged with Chapter 5 Study Guide For Content Mastery Answer Key Chemistry
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chapter 5 Study Guide For Content Mastery Answer Key Chemistry
9. Balancing eBooks and Physical Books Chapter 5 Study Guide For Content Mastery Answer Key Chemistry
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chapter 5 Study Guide For Content Mastery Answer Key Chemistry
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Chapter 5 Study Guide For Content Mastery Answer Key Chemistry
  - Setting Reading Goals Chapter 5 Study Guide For Content Mastery Answer Key Chemistry
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chapter 5 Study Guide For Content Mastery Answer Key Chemistry
  - Fact-Checking eBook Content of Chapter 5 Study Guide For Content Mastery Answer Key Chemistry
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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