

0620 Chemistry Past Papers (Year 2022) – Paper 2 MCQs

Feb/Mar 2022 – 22

1. Which gas has the fastest rate of diffusion?
- A H_2 B CH_4 C CO_2 D SO_2
2. In which state does 1 dm^3 of methane contain the most particles?
- A gas at 100°C
B gas at room temperature
C liquid
D solid
3. Which statement explains why isotopes of the same element have the same chemical properties?
- A They have the same electronic structure.
B They have the same relative mass.
C They have the same nucleon number.
D They have the same proton number.

4. The electronic structures of atoms P and Q are shown.



P and Q form an ionic compound.

What is the formula of the compound?

- A PQ B P_2Q C P_2Q_3 D PQ_3
5. Fermentation of sugar produces a mixture of ethanol solution and solid yeast.
- How is the solid yeast removed from the mixture?
- A crystallisation
B distillation
C filtration
D fractional distillation
6. Compound X contains carbon, hydrogen and oxygen only.
- By mass, it contains 26.7% carbon and 2.2% hydrogen.
- What is the empirical formula of X?

- A CHO B C_2HO C CH_2O D CHO_2

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