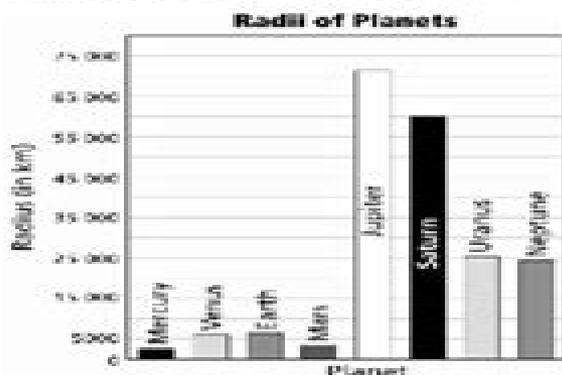


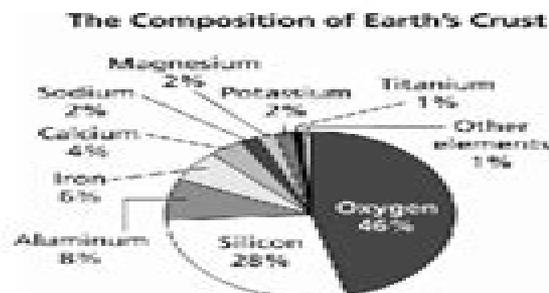
CHAPTER 2 Analyzing Data

1. A sample of aluminum is placed in a 25-mL graduated cylinder containing 10.0 mL of water. The level of water rises to 13.0 mL. Aluminum has a density of 2.7 g/mL. Calculate the mass of the sample.
2. Saturn is about 1,429,000 km from the Sun. How many meters is Saturn from the Sun? Write your answer in scientific notation.
3. Use the graph to answer the questions.



- a. What kind of graph is this?
- b. What are the variables?
- c. According to the graph, which has a larger radius, Neptune or Uranus?
- d. According to the graph, what is the radius of Saturn?
- e. Convert the radius of Saturn to meters. Write your answer in scientific notation.

4. Look at the graph below. Then answer the questions.



- a. What kind of graph is this?
 - b. According to the graph, which element is most abundant in Earth's crust?
 - c. According to the graph, what percent of Earth's crust is made up of titanium? Of calcium?
5. You place a 26.95-g piece of gold in a 10-mL graduated cylinder. The level of the water rises 1.50 mL. What is the density of gold? You know that silver has a density of 10.5 g/cm³. What mass of silver will raise the level of the water in the graduated cylinder 1.50 mL?
 6. Convert 55 miles per hour to kilometers per hour. How many kilometers/second is 55 miles per hour? (1 mile = 1.6 km)
 7. Convert the following data to scientific notation.
 - a. 166,000,000,000,00 m²
 - b. 3347 m
 - c. 484 L

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Hussin A.Rothana



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