

Chapter Test B

Chapter: Chemical Equations and Reactions

Match a name space (possibly writing the name of the react or product molecule) completed with reactants/products atoms/molecules.

- The product(s) of a chemical reaction will be represented by a double arrow pointing towards the right. The substance(s) on the left side of the reaction is/are
 - reactant(s)**
 - unbalanced equation
 - soluble
- An ionic chemical equation is represented by adding the
 - charges**
 - charges of the particles
 - number of particles
- An ionic equation that is written as aq indicates the substance is
 - solid
 - dissolved in water**
 - liquid
 - volatile
- The activity for balancing chemical equations is called balancing a
 - chemical formula using mass balancing for the particles participating
 - chemical formula
 - redox chemical reaction
 - chemical formula using mass balancing for the particles participating**
- The coefficient for chemical equations
 - indicates the number of moles of each substance that react**
 - shows the number of grams of each substance that react
 - shows the number moles of the substances
 - shows the relative proportions of each atom
- If you're to balance the mass of H_2O with water molecules, then
 - 1 water molecule has 1 H and 1 oxygen**
 - 1 water molecule has 2 H and 1 oxygen
 - 1 water molecule has 1 H and 1 oxygen
 - 1 water molecule has 2 H and 2 oxygen

Chapter 8 Test B

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