

Chapter Test A

Chapter: Chemical Equations and Reactions

Do the work provided below the letter of the item or phrase that best completes each statement or best answers each question.

1. The mass number of an atom is the sum of the number of _____ and the number of _____ in the nucleus of the atom.
 - a. electrons; neutrons
 - b. protons; neutrons
 - c. protons; electrons
 - d. neutrons; electrons
2. An element is described by a chemical equation involving n moles of a substance reacting with 1 mole of oxygen.
 - a. oxidized by n moles of oxygen
 - b. oxidized by $2n$ moles of oxygen
 - c. reduced by n moles of oxygen
 - d. reduced by $2n$ moles of oxygen
3. Which of the following relationships is always true for a chemical reaction?
 - a. The products of a reaction are more stable than the reactants.
 - b. All of the reactants are used.
 - c. The products of a reaction are more stable than the reactants.
 - d. The reactants of a reaction are more stable than the products.
4. There is a negative enthalpy change for a chemical reaction if the reaction
 - a. is exothermic.
 - b. is endothermic.
 - c. produces more products than reactants.
 - d. all of the above.
5. The enthalpy of a chemical reaction is the
 - a. heat of reaction.
 - b. heat of formation.
 - c. heat of combustion.
 - d. heat of fusion.

Chapter 9 Modern Chemistry Test

SA Adler



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