

Thermodynamic Equilibrium - 1 (Application of Equilibrium)

Thermodynamic Equilibrium

Thermodynamic Equilibrium states that both conditions are necessary to produce the equilibrium:

1. The temperature of system and surroundings (thermosphere) must be the same (either)
- Equal ✓
 - Equal
 - Not at the same
2. The free enthalpy of the system must be the same (either both are)
- Equal ✓
 - Equal
 - Not at
3. There must be no net change in the free enthalpy of the system when a small amount of matter is transferred
- Different
 - Different
 - Different ✓
 - None at the same
4. There must be no net change in the free enthalpy of the system when a small amount of matter is transferred
- Different
 - Different
 - Equal ✓
 - Not at the same
5. There must be no net change in the free enthalpy of the system when a small amount of matter is transferred
- Equal (temperature of system)
 - Equal both (temperature)
 - Equal both
 - Equal (temperature of surroundings)
6. There must be no net change in the free enthalpy of the system when a small amount of matter is transferred
- Equal ✓
 - Equal
 - Equal
 - Equal
7. There must be no net change in the free enthalpy of the system when a small amount of matter is transferred
- Equal
 - Different
 - Different ✓
 - Different
8. There must be no net change in the free enthalpy of the system when a small amount of matter is transferred
- Equal (temperature of system)
 - Equal both (temperature)
 - Equal both (temperature)
 - Equal both (temperature) ✓

Chapter Test Thermal Energy Answers

Albert A Gayle



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