

## CHAPTER 11 ASSESSMENT CHEMISTRY ANSWERS

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#### ANSWER 11.1: ASSESSMENT CHEMISTRY ANSWERS

1. The first two reactions are not balanced. The balanced chemical equations are:

$2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$  and  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$  (Note: The original text has a typo in the second equation, it should be  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ ).

2. The balanced chemical equation is:  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$

3. The balanced chemical equation is:  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$  (Note: The original text has a typo in the equation, it should be  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ ).

4. The balanced chemical equation is:  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$  (Note: The original text has a typo in the equation, it should be  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ ).

[The following text is heavily redacted and illegible. It appears to contain several paragraphs of text, possibly including chemical equations and explanations, but the content cannot be discerned.]

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**WJ Hussar**



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