

# 17

## THERMOCHEMISTRY

Reaction that releases heat is exothermic  
 Reaction that absorbs heat is endothermic  
 Exothermic:  $\Delta H < 0$   
 Endothermic:  $\Delta H > 0$

### Practice Problems

1. How much heat is released when 100.0 g of water is cooled from 25.0°C to 5.0°C?

### SECTION 17.1 THE FLOW OF ENERGY—HEAT AND WORK

1. How much heat is released when 100.0 g of water is cooled from 25.0°C to 5.0°C?

2. How much heat is released when 100.0 g of water is cooled from 25.0°C to 5.0°C?

3. How much heat is released when 100.0 g of water is cooled from 25.0°C to 5.0°C?

4. How much heat is released when 100.0 g of water is cooled from 25.0°C to 5.0°C?

5. How much heat is released when 100.0 g of water is cooled from 25.0°C to 5.0°C?

6. How much heat is released when 100.0 g of water is cooled from 25.0°C to 5.0°C?

7. How much heat is released when 100.0 g of water is cooled from 25.0°C to 5.0°C?

8. How much heat is released when 100.0 g of water is cooled from 25.0°C to 5.0°C?

9. How much heat is released when 100.0 g of water is cooled from 25.0°C to 5.0°C?

10. How much heat is released when 100.0 g of water is cooled from 25.0°C to 5.0°C?

### SECTION 17.2 MEASURING AND EXPRESSING ENTHALPY CHANGES

1. How much heat is released when 100.0 g of water is cooled from 25.0°C to 5.0°C?

2. How much heat is released when 100.0 g of water is cooled from 25.0°C to 5.0°C?

3. How much heat is released when 100.0 g of water is cooled from 25.0°C to 5.0°C?

4. How much heat is released when 100.0 g of water is cooled from 25.0°C to 5.0°C?

5. How much heat is released when 100.0 g of water is cooled from 25.0°C to 5.0°C?

6. How much heat is released when 100.0 g of water is cooled from 25.0°C to 5.0°C?

7. How much heat is released when 100.0 g of water is cooled from 25.0°C to 5.0°C?

8. How much heat is released when 100.0 g of water is cooled from 25.0°C to 5.0°C?

9. How much heat is released when 100.0 g of water is cooled from 25.0°C to 5.0°C?

10. How much heat is released when 100.0 g of water is cooled from 25.0°C to 5.0°C?

### SECTION 17.3 HEAT IN CHANGES OF STATE

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**S Ben Porath**



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