

Chapter 25 Nuclear Chemistry Practice Test

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- ____ 1. What does the 4 in ${}^4_2\text{He}$ represent?
a. the mass number
b. the atomic number
c. the number of protons
d. the number of neutrons
- ____ 2. What does the 101 in ${}^{256}_{101}\text{Md}$ represent?
a. the mass number
b. the atomic number
c. the nuclide number
d. the number of neutrons
- ____ 3. Balance the following equation: ${}^{226}_{88}\text{Rn} \rightarrow {}^{226}_{86}\text{Rn} + \text{_____}$
a. ${}^4_2\text{He}$
b. ${}^1_0e^-$
c. ${}^1_1\text{H}$
d. ${}^0_{-1}e^-$
- ____ 4. Balance the following equation: ${}^{239}_{93}\text{Np} \rightarrow \text{_____} + {}^0_{-1}e^-$
a. ${}^{239}_{90}\text{Th}$
b. ${}^{239}_{92}\text{U}$
c. ${}^{239}_{94}\text{Pu}$
d. ${}^{238}_{94}\text{Pu}$
- ____ 5. The spontaneous disintegration of a nucleus into a slightly lighter and more stable nucleus, accompanied by emission of particles, electromagnetic radiation, or both, is
a. nuclear fusion.
b. nuclear radiation.
c. radioactive decay.
d. nuclear fission.
- ____ 6. Which of the following forms of radiation has the greatest penetrating power?
a. alpha particles
b. beta particles
c. gamma rays
d. positrons
- ____ 7. Alpha particles are
a. electrons.
b. helium nuclei.
c. electromagnetic waves.
d. neutrons.
- ____ 8. Beta particles are
a. electrons.
b. helium nuclei.
c. electromagnetic waves.
d. neutrons.
- ____ 9. Which of the illustrations below represents a fission reaction?

Chapter 25 Nuclear Chemistry D Reading Answers

Silberberg



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Chemistry McGraw-Hill Staff, 2001-07 53 Previous Years Iit-Jee Main and Advanced Chapter-Wise Solved Papers 1970-2022 Chemistry Dr. K G Ojha, Dr. Sunita, 2023-03-25 The New 2023 Edition of IIT JEE Main Advanced Chemistry is designed to present a whole package of Chemistry study preparation sufficing the requirements of the aspirants who are preparing for the upcoming exam Highlights of the Book Exam Patterns for JEE Main and Advanced included An Analysis of IIT JEE included Concepts are explained in detail Chapters are compiled with Previous Years Questions Answers to Questions included with Explanations Presence of accurate Figures and Tables Five sets of Mock Tests are also included at the end Based on the pattern of NCERT Books 53 Years of IIT JEE Chapter wise Topic wise Solved Papers Chemistry 1970 2022 with Value Added Notes covers the whole syllabus distributing in 30 Chapters The book comprises chapters such as Stoichiometry Solutions Atomic Structure Redox Electrochemistry Alcohols Phenols and Ethers Biomolecules Analytical Chemistry and Experimental Skills and so on This book serves to be a suitable Study Guide for the aspirants with focus on Qualitative Preparation and Systematic understanding of the Syllabus and Examination Level With provision for self assessment in Mock Tests this book stands beneficial in imprinting concepts in the mind **Whitaker's Cumulative Book List**, 1969

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