

Chapter 1
Properties of Matter

Properties of Matter

QUESTIONS

Are you able distinguish, how these questions are asked?

- Which one(s) among the following is/are physical property?
- Which one(s) among the following is/are chemical property?

What are Physical Properties?

Such as color, taste, smell, texture, boiling point etc. are called physical properties of the substance.



If physical properties of a substance are observed without changing the identity of the substance, then these are called physical properties of the substance. For example, you can identify an object's color, taste, shape, etc. or weight, etc. without the change of its identity. These are determined by the properties of the substance. The shape remains the same.

Many of the physical properties of substances like melting point, boiling point, etc. are called physical properties. These properties include density, hardness, magnetism, and the ability to conduct heat and electricity.

Some physical properties depend on the state of a substance. For example, the boiling point of water is 100°C at sea level. This is because the boiling point of water is 100°C at sea level. Other physical properties are called intensive properties. They do not depend on the amount of the substance. They are called intensive properties. They are called intensive properties because they are not affected by the amount of the substance. They are called intensive properties because they are not affected by the amount of the substance.

LEARNING OBJECTIVES

After this chapter, you should be able to:

- 1. Distinguish between physical and chemical changes.
- 2. Identify physical and chemical properties of matter.

LEARNING CHECKS

1. Identify the physical and chemical properties of the following substances.

EXERCISES

1. Identify the physical and chemical properties of the following substances.

Chapter 2 Properties Matter Wordwise Answers

LP Steffe



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