

UNIT 10: PERIODIC TABLE **PERIODIC TRENDS** **PERIODIC TRENDS**

Section 10.1 (continued)

In your notebook, read aloud the student periodicals table.

Copy the letterheads in the first two columns below from the periodic table to complete the table on the right.

Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Group Number	_____
Period Number	_____

Put each item in Column A, under the letter of the matching item in Column B.

Column A

Column B

- | | |
|---|----------------------------|
| _____ 18. A column in the periodic table | a. metals |
| _____ 19. A row in the periodic table | b. group |
| _____ 20. Group 1 elements | c. period |
| _____ 21. Elements that are shiny and conduct electricity | d. representative elements |
| _____ 22. Group 17 elements | e. transition elements |

In the space on the left, write each of the statements in terms of the statement in bold. Change the underlined word or phrase to make it true.

- _____ 23. There are two representative elements of platinum.
- _____ 24. Most nonmetals form diatomic molecules in the periodic table (or naturally).
- _____ 25. Group 1A elements except for hydrogen are located on the left side.
- _____ 26. Group 1A elements are the alkali metal metals.
- _____ 27. Group 1A elements are highly reactive nonmetals known as halogens.
- _____ 28. Group 1A elements are (or) nonmetallic elements located on representative metals.
- _____ 29. Alkali metals have properties of both metals and more reactive metals.

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Scott C. Dulebohn



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