

## Chapter 6 Thermochemistry

### Homework Problems

- The enthalpy change for the fermentation (without oxygen,  $O_2$  gas) reaction of glucose to alcohol and carbon dioxide is  $-67$  kJ.
  - Write a chemical equation for the fermentation reaction and show how energy is involved in the reaction.
  - Is the reaction exothermic or endothermic?
  - Is energy, in the form of heat, absorbed or evolved as the reaction occurs?
- The specific heat capacity of graphite is  $0.71$  J/°C·g. Calculate the energy required to:
  - Raise the temperature of  $1.0$  mol graphite by  $1.0$ °C.
  - Raise the temperature of  $850$  g graphite by  $150$ °C.
  - Raise the temperature of  $75$  kg graphite from  $294$  K to  $348$  K.
- In a coffee-cup calorimeter,  $100.0$  mL of  $1.0$  M NaOH and  $100.0$  mL of  $1.0$  M HCl are mixed. Both solutions were originally at  $24.6$ °C. After the reaction, the final temperature is  $31.3$ °C. Assume that all the solutions have a density of  $1.0$  g/mL and a specific heat capacity of  $4.18$  J/g°C. Calculate the enthalpy change for the neutralization of HCl by NaOH. Assume that no heat is lost to the surroundings or to the calorimeter.
- Camphor ( $C_{15}H_{26}O$ ) has an energy of combustion of  $-5903.6$  kJ/mol. A sample of camphor with a mass of  $0.1204$  g is burned in a bomb calorimeter, the temperature increases by  $2.28$ °C. Calculate the heat capacity of the calorimeter.  
A  $0.1964$  g sample of quinone ( $C_6H_4O_2$ ) is burned in a bomb calorimeter that has a heat capacity of  $1.56$  kJ/°C. The temperature of the calorimeter increases by  $3.2$ °C. Calculate the energy of combustion of quinone per gram and per mole.

# Chapter 6 Thermochemistry Homework Problems

**Amyand David Buckingham**



## **Chapter 6 Thermochemistry Homework Problems:**

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**H. A. Skinner** Amyand David Buckingham,1972      **Must Know High School Chemistry, Second Edition** Mary Millhollon,Richard H. Langley,2022-06-12 A unique and effective way to learn Chemistry updated with the latest instruction and review Must Know High School Chemistry provides a fresh approach to learning As part of our Must Know series this new edition makes sure what you really need to know is clear up front Rather than starting with goals to be met chapters begin by telling you the most important concepts about the topic at hand and then show you exactly how these concepts help you accomplish your goals Written by expert chemistry educators Must Know High School Chemistry Second Edition provides updated lesson content and useful examples to help clarify each topic Every chapter closes with reinforcing exercises to get you the practice you need to gain confidence New features to this edition focus on extra support and helping you avoid common mistakes In the end you get everything you need to build your chemistry skills quickly and painlessly Features More than 350 practice questions that parallel what you will find in your classwork and on exams Bonus app that includes 100 flashcards to reinforce concepts Extra Help and Easy Mistake features put the emphasis on how to improve and what pitfalls to avoid Chemistry topics aligned to national and state educational standards Practical examples throughout and an answer key with explanations make sure you understand the topics Conversational writing style and informative IRL In Real Life and BTW By the Way sidebars A special section for teachers with tips and strategies on teaching the material and content specific links and resources      **Chemical Principles** William L. Masterton,Emil J. Slowinski,Conrad L. Stanitski,1981      General Chemistry Darrell D. Ebbing,1999      **General Chemistry Key Concise Review Guide Sixth Edition** Darrell D. Ebbing,1999      Engineering News ,1906      **Complete Preparation for the MCAT** Williams & Wilkins Review,1998-04 Here is the most respected test prep book for the Medical College Admission Test you can buy featuring an active learning approach for a better understanding of the exam s content and a better chance for success Unique to this guide are coverage of all recent changes in the MCAT plus a step by step plan for sharpening cognitive skills developing problem solving skills and critical thinking This thorough guide replaces expensive test preparation courses while giving students exactly what they need to get ready for the MCAT

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    - Chapter 5: Conclusion
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