

## CHEMISTRY REVIEW: KEY CHAPTER 9 – CHEMICAL NAMES AND FORMULAS

- In a Stock name such as iron (II) sulfate, the roman numeral tells us **C**
  - How many atoms of Fe are in one formula unit
  - How many sulfate ions can be attached to the iron atom
  - The charge on each Fe ion
  - The total positive charge of the formula unit
- The result of changing a subscript in a correctly written chemical formula is to **c**
  - Change the number of moles represented in the formula
  - Change the charges on the other ions in the compound
  - Change the formula so that it no longer represents the compound
  - Have no effect on the formula
- Derive a generalization for when an acid name will end in the suffix **-ic** or **-ous**  
**If ending is -ic, the anion ended in -ate; if ending is -ous, the anion ended in -ite**
- Derive a generalization for when an acid name will begin with the prefix **hydro-**  
**If prefix is hydro-, the anion ended in -ide. This is most likely a monatomic ion.**
- Fill in the blanks in the table below

Compound Name	Formula
Aluminum sulfide	$Al_2S_3$
Aluminum sulfate	$Al_2(SO_4)_3$
Lead (II) chloride	$PbCl_2$
Ammonium phosphate	$(NH_4)_3PO_4$
Hydrobromic acid	$HBr$
Dinitrogen pentoxide	$N_2O_5$
Copper (II) carbonate	$CuCO_3$
Sodium sulfide	$Na_2S$
Ammonium phosphate	$(NH_4)_3PO_4$
Tin (IV) sulfide	$Sn(SO_4)_2$
Nitrous acid	$HNO_2$
Magnesium perchlorate	$Mg(ClO_4)_2$
Iron (II) nitrate	$Fe(NO_3)_2$
Iron (III) nitrate	$Fe(NO_3)_3$
Cobalt (II) oxide	$CoO$
Sulfur trioxide	$SO_3$
Boron pentafluoride	$BF_5$
Tetraphosphorus octasulfide	$P_4S_8$
Tetraoxide nonoxide	$TeO_9$

- Name the following compound and identify as molecular, ionic or acid
  - $Ca(HCO_3)_2$  calcium bicarbonate - **I**
  - $Fe(ClO_2)_3$  iron (III) chlorite - **I**
  - $Fe(ClO_4)_2$  iron (II) chlorate - **I**
  - $N_2O_5$  dinitrogen pentoxide - **M**

# Chapter 9 Chemical Names And Formulas Answers

**Disha Experts**



## **Chapter 9 Chemical Names And Formulas Answers:**

## Reviewing **Chapter 9 Chemical Names And Formulas Answers**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Chapter 9 Chemical Names And Formulas Answers**," an enthralling opus penned by a highly acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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