

## General Chemistry 11

### Activity Sheet 1 for Quarter 1 Week 3 (Paired/Individual Activity)

(TUESDAY)

**Reminders: Accomplish the activities every Tuesday and Thursday during your schedule of Chemistry class. Submit all answers in the google classwork.**

#### Activity 1. What a Molar Mass!

**Activity 1.a. Calculate the molar mass of the following compounds:**

Compound	Solution	Molar Mass
CO <sub>2</sub>	$  \begin{array}{r}  \text{C} = 12 \qquad 12 \times 1 = 12 \\  \text{O} = 16 \qquad 16 \times 2 = 32 \\  \hline  44\text{g}  \end{array}  $	44g/mol
Fe <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	$  \begin{array}{r}  \text{Fe} = 56 \qquad 56 \times 3 = 168 \\  \text{P} = 31 \qquad 31 \times 2 = 62 \\  \text{O} = 16 \qquad 16 \times 4 = 64 \times 2 = 128 \\  \hline  358\text{g}  \end{array}  $	358g/mol
CH <sub>3</sub> COOH	$  \begin{array}{r}  \text{C} = 12 \qquad 12 \times 2 = 24 \\  \text{H} = 1 \qquad 1 \times 3 + 1 = 4 \\  \text{O} = 16 \qquad 16 \times 2 = 32 \\  \hline  60\text{g}  \end{array}  $	60g/mol
H <sub>2</sub> O	$  \begin{array}{r}  \text{H} = 1 \qquad 1 \times 2 = 2 \\  \text{O} = 16 \qquad 16 \times 1 = 16 \\  \hline  18\text{g}  \end{array}  $	18g/mol
CaSO <sub>4</sub>	$  \begin{array}{r}  \text{Ca} = 40 \qquad 40 \times 1 = 40 \\  \text{S} = 32 \qquad 32 \times 1 = 32 \\  \text{O} = 16 \qquad 16 \times 4 = 64 \\  \hline  136\text{g}  \end{array}  $	136g/mol

**Activity 1.b. Direction: Identify whether the following is an empirical or molecular formula:**

- |   |  |
|---|--|
| a. CH <u>Empirical Formula</u>                | d. CH <sub>2</sub> O <u>Molecular Formula</u>                            |
| b. CO <sub>2</sub> <u>Empirical Formula</u>   | e. C <sub>3</sub> H <sub>4</sub> O <sub>3</sub> <u>Empirical Formula</u> |
| c. AlCl <sub>3</sub> <u>Molecular Formula</u> |  |

# Chem 1 2 Weebly

**CH Cherryholmes**



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## **Table of Contents Chem 1 2 Weebly**

1. Understanding the eBook Chem 1 2 Weebly
  - The Rise of Digital Reading Chem 1 2 Weebly
  - Advantages of eBooks Over Traditional Books
2. Identifying Chem 1 2 Weebly
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chem 1 2 Weebly
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chem 1 2 Weebly
  - Personalized Recommendations
  - Chem 1 2 Weebly User Reviews and Ratings
  - Chem 1 2 Weebly and Bestseller Lists
5. Accessing Chem 1 2 Weebly Free and Paid eBooks

- Chem 1 2 Weebly Public Domain eBooks
  - Chem 1 2 Weebly eBook Subscription Services
  - Chem 1 2 Weebly Budget-Friendly Options
6. Navigating Chem 1 2 Weebly eBook Formats
    - ePub, PDF, MOBI, and More
    - Chem 1 2 Weebly Compatibility with Devices
    - Chem 1 2 Weebly Enhanced eBook Features
  7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Chem 1 2 Weebly
    - Highlighting and Note-Taking Chem 1 2 Weebly
    - Interactive Elements Chem 1 2 Weebly
  8. Staying Engaged with Chem 1 2 Weebly
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Chem 1 2 Weebly
  9. Balancing eBooks and Physical Books Chem 1 2 Weebly
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Chem 1 2 Weebly
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Chem 1 2 Weebly
    - Setting Reading Goals Chem 1 2 Weebly
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Chem 1 2 Weebly
    - Fact-Checking eBook Content of Chem 1 2 Weebly
    - Distinguishing Credible Sources
  13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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