

You must show all your calculation set-ups to receive full credit. You must report all answers to the correct # of significant figures (sig. figs.).

1. One mole of any gas at STP occupies a volume of

- a. 22.4 L
 b. 1.00 L
 c. 24.4 L
 d. 24.4 mL
 e. 1.00 mL

$$\frac{85}{100} + 9 = \frac{94}{100} \text{ great!}$$

2. The density of a gas is reported in units of

- a. g/cm³
 b. atm
 c. mm Hg
 d. g/mL
 e. g/L

3. Which color of visible light will have the longest wavelength of radiation?

- a. red
 b. yellow
 c. blue
 d. violet
 e. orange

VR to GR

4. For an exothermic reaction, ΔH is always

- a. positive
 b. negative
 c. zero
 d. small

$$\Delta H = \frac{-q}{n}$$

5. When calculating the root mean square velocity of a sample of chlorine gas, we should use which of the following for the molar mass?

- a. 35.45 g/mol
 b. 0.03545 kg/mol
 c. 70.90 g/mol
 d. 0.07090 kg/mol

$$r_{ms} = \sqrt{\frac{3RT}{M}} \rightarrow \text{kg/mol } (35.45 \text{ g}) \frac{\text{kg}}{1000 \text{ g}} = 0.03545 \text{ kg}$$

6. The symbol (*) in ΔH^* represents which of the following conditions?

- a. 1 atm, 0°C → ΔP
 b. 1 atm, 298K → ΔE
 c. 1.013 atm, 25°C
 d. 1.013 atm, 273K

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