

7. In an exothermic reaction, is enthalpy change positive or negative?

.....

8. In an endothermic reaction, is enthalpy change positive or negative?

.....

9. When hydrochloric acid reacts with ammonium hydroxide in a beaker, the temperature goes up.



$$\Delta H = -53.4 \text{ kJ/mol}$$

Complete the energy profile diagram and state whether the reaction is endothermic or exothermic, explain your answer.



10. What are the units for enthalpy change,  $\Delta H$

.....

You may like to watch this lesson on the different types of Chemical reactions:



Energy Changes in  
Chemical reactions  
for KS3 Science

# Chemical Reactions And Energy Worksheet Answers

**D Siedentop**



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