



Equilibrium

- **Equilibrium** is a state where the net force on an object is zero. It can be static or dynamic.
- **Static Equilibrium**: The object is at rest.
 - $\sum F_x = 0$ and $\sum F_y = 0$
 - $\sum \tau = 0$
- **Dynamic Equilibrium**: The object is moving with constant velocity.
 - $\sum F_x = 0$ and $\sum F_y = 0$
 - $\sum \tau = 0$
- **Center of Mass**: The point where the mass of the object is concentrated.
 - $\sum \tau = 0$ about the center of mass.
- **Stable Equilibrium**: The object returns to its original position after a small displacement.
- **Unstable Equilibrium**: The object moves away from its original position after a small displacement.
- **Neutral Equilibrium**: The object remains in its original position after a small displacement.



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Nuclear Chemistry

- **Nuclear Chemistry** is the study of the changes in the nucleus of an atom. It involves the emission of particles and energy.
- **Radioactivity**: The process by which an unstable nucleus emits particles and energy.
 - α decay: Emission of a helium nucleus (^4_2He).
 - β decay: Emission of an electron ($^0_{-1}\text{e}$) or a positron ($^0_{+1}\text{e}$).
 - γ decay: Emission of a high-energy photon (γ).
- **Half-life**: The time taken for half of the radioactive substance to decay.
- **Types of Processes**
 - **Radioactive Decay**: The process by which an unstable nucleus emits particles and energy.
 - **Nuclear Fission**: The process by which a heavy nucleus splits into two lighter nuclei, releasing energy.
 - **Nuclear Fusion**: The process by which two light nuclei combine to form a heavier nucleus, releasing energy.
- **Applications**
 - **Medical**: Radioactive isotopes are used in diagnosis and treatment.
 - **Industrial**: Radioactive isotopes are used in various industries.
 - **Energy**: Nuclear fission and fusion are used to generate energy.



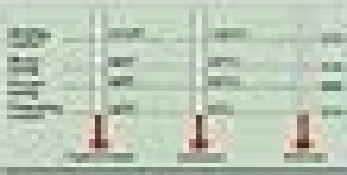
Measurement & Units

- **Measurement** is the process of comparing a quantity with a standard unit.
- **Units** are the standards used for measurement.
 - **SI Units**: International System of Units.
 - **Base Units**: The seven fundamental units.
 - **Derived Units**: Units derived from the base units.
- **Significant Figures**: The digits in a number that carry meaning contributing to its measurement accuracy.
- **Dimensional Analysis**: The process of checking the units of a calculation to ensure they are correct.
- **Accuracy**: The closeness of a measurement to the true value.
- **Precision**: The closeness of repeated measurements to each other.



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Mark Steyvers



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