

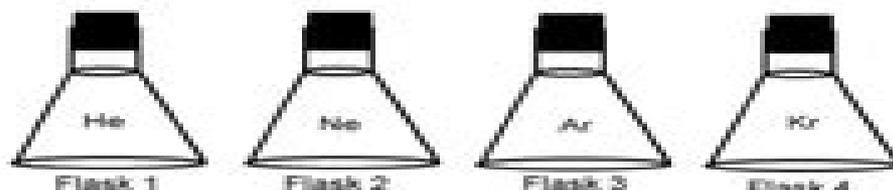
Gases Test SCH3U

Name: _____

$\frac{\text{K}}{15}$ $\frac{\text{T}}{9}$ $\frac{\text{C}}{5}$ $\frac{\text{A}}{10}$ **Assessment of Learning (Product)**

Multiple Choice. Circle the most correct answer. (15K)

- Convert the pressure 1.80 kPa to atm.
 A) 2 atm B) 115 atm C) 132 atm D) 245 atm E) None of the above
- What happens to the temperature of a gas when it is expanded quickly?
 A) Temperature increases B) Temperature does not change C) Temperature decreases



- If all of the following flasks are the same size, at the same temperature, and contain the same number of molecules, in which flask will the pressure be highest?
 A) All have the same pressure B) Flask 1 C) Flask 2
 D) Flask 3 E) Flask 4

- Which diagram best represents a gas in a closed container?



- According to the kinetic molecular theory, the Kelvin temperature of an ideal gas is proportional to which one of the following.

A) acceleration B) momentum C) kinetic energy D) potential energy

- A gas is _____ and assumes _____ of its container whereas a liquid is _____ and assumes _____ of its container.

A) compressible, the volume and shape, not compressible, the shape of a portion
 B) compressible, the shape, not compressible, the volume and shape
 C) compressible, the volume and shape, compressible, the volume
 D) condensed, the volume and shape, condensed, the volume and shape
 E) condensed, the shape, compressible, the volume and shape

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J Dewey



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