

GOLD
Chemistry

Grade 12 Chemistry New Course

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Chapter-2

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Enthalpy Cycle Calculations

Lesson - 17

September 1, 2023

$$\begin{aligned} & \Delta H + \Delta H \\ & = \text{enthalpy change for} + \\ & \text{bonds broken} \\ & = (E(N \equiv N)) + 3(E(H-H)) + \\ & = 945 + 3 \times (436) + \\ & = 2253 + (-2) \\ & = -93 \text{ kJ} \end{aligned}$$



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