

Chemical Reactions

Intro. to Chemical Reactions

Name: _____
 Date: _____
 Class: _____

Electrolysis Reaction

1. Write the chemical formula following compounds:

a) hydrogen gas: H_2 b) sodium phosphate: Na_3PO_4
 c) magnesium phosphate: $Mg_3(PO_4)_2$ d) sodium sulfate: Na_2SO_4

2. Write the correct formula following compounds:

a) $Ca(NO_3)_2$: calcium nitrate b) $PbCl_2$: lead(II) chloride
 c) $CaHCO_3$: calcium bicarbonate d) NaF : sodium fluoride

Chemical Equations: What they look like

Here is what a chemical reaction looks like when it is written:



There are examples that you are going to describe what happens during:

Chemical Equations react with **stoichiometry** and **stoichiometry** and **stoichiometry**

There are also some stoichiometry concepts like **stoichiometry** and **stoichiometry**. Here are a few more:

Critical Thinking Questions

1. Write the following reaction, with the correct gas laws with it:



Chemical Equations react with **stoichiometry** and **stoichiometry** and **stoichiometry**

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