

Chapter 10 Chemical Quantities Review

Part 1: Vocabulary

Label the structure corresponding to the term in the vocabulary list.

- | | |
|-------------------------|---------------------------|
| _____ mole | _____ percent composition |
| _____ empirical formula | _____ empirical formula |
| _____ molar mass | _____ molar mass |
| _____ molar mass | _____ molar mass |
10. _____ is a description of the relative amounts of each element in a compound. _____ percent composition
11. _____ is the lowest whole number ratio of the atoms of the elements in a compound. _____ empirical formula
12. _____ is the mass of one mole of any gas at STP. _____ molar mass
13. _____ is the mass of one mole of a compound. _____ molar mass
14. _____ is the amount representing 6.02×10^{23} representative particles of a substance. _____ mole
15. _____ is 6.02×10^{23} particles. _____ Avogadro's number
16. _____ is the mass of one mole of a substance at STP. _____ molar mass
17. _____ is the mass of one mole of a gas at STP. _____ molar mass
18. _____ is the mass of one mole of a gas at STP. _____ molar mass

Part 2: Section Concept Review

10.1 The Mole: A Fundamental

Complete the following practice problems using your knowledge of the terms and your understanding of the concepts involved with each of the terms. Write your answers on separate sheets of paper.

1. _____ is the mass of one mole of a substance. _____ molar mass

2. _____ is the mass of one mole of a substance. _____ molar mass

3. _____ is the mass of one mole of a substance. _____ molar mass

4. _____ is the mass of one mole of a substance. _____ molar mass

Practice Problems

1. _____ is the mass of one mole of a substance. _____ molar mass

2. _____ is the mass of one mole of a substance. _____ molar mass

3. _____ is the mass of one mole of a substance. _____ molar mass

4. _____ is the mass of one mole of a substance. _____ molar mass

5. _____ is the mass of one mole of a substance. _____ molar mass

6. _____ is the mass of one mole of a substance. _____ molar mass

Answers

Chapter 10 Chemical Quantities Guided Reading Answer Key

TD Snyder



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