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Chapter 17 & 19 Study Guide

Key Terms

Directions: Define each of the following key terms from chapters seventeen and nineteen on a separate sheet of paper.

- **Atom** - the smallest piece of matter that still has the properties of the element
- **atomic number** – number of protons in an atom's nucleus
- **atomic mass** – average mass of all an element's isotopes
- **ductile** – capable of being drawn into wire
- **electron** – subatomic particle with a negative charge
- **electron cloud** – area outside the nucleus in which electrons can be found.
- **Electron dot diagram** – uses the element's symbol and dots to represent the valence electrons in an element.
- **energy level** – areas within the electron cloud where electrons are most likely to be found.
- **Family** – vertical columns of the periodic table
- **Group** - vertical columns of the periodic table
- **half-life** – the amount of time it takes half the sample of a radioactive isotope to decay.
- **Isotope** – atoms of the same element that have different numbers of neutrons.
- **Lustrous** – having a shiny appearance
- **Malleable** – can be hammered flat or rolled into sheets
- **mass number** – sum of protons and neutrons in the nucleus of an atom.
- **Metal** – element that typically is hard, shiny, solid, malleable, and is a good conductor of heat and electricity.
- **Neutron** – neutrally charged subatomic particle.
- **Nonmetal** - element that typically is brittle solid or gas, not shiny, not malleable, and is a poor conductor of heat and electricity.
- **Nucleus** – center of the atom
- **Period** – horizontal row of the periodic table.
- **periodic table** – organized list of all known elements.
- **Proton** – positively charged subatomic particle
- **transition metal** – Elements found in families 3-12 of the periodic table.
- **transuranium element** – an element containing more than 92 protons in its nucleus
- **quark** – particle believed to make up protons and neutrons.
- **valence electron** – electrons found in the outermost energy level of an atom.

Chapter 17 Study Guide Answers

JG Myers



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Chapter 17 Study Guide Answers Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Chapter 17 Study Guide Answers**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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