

**Chapter 1**  
**Properties of Matter**

**Properties of Matter**

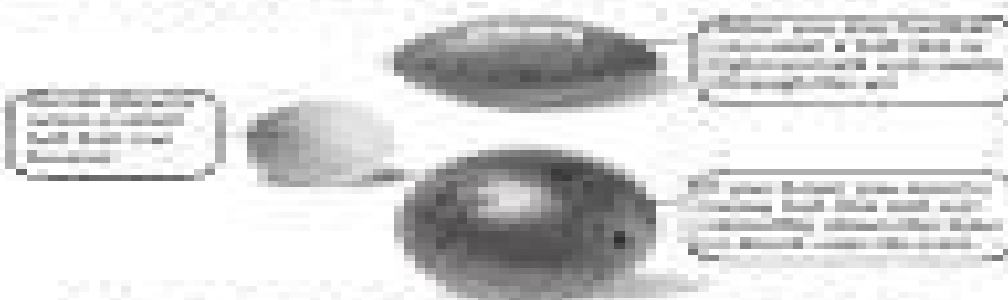
**QUESTIONS**

Are you able distinguish, how these questions are asked?

- Why separating, mixture and showing physical properties?
- Why can't separating, mixture by its chemical properties?

**What are Physical Properties?**

Start with the title in the chapter before you start to read and understand the meaning of these common properties.



If physical properties of substances are observed without changing the identity of the substance, these are called physical properties. For example, you can identify an object's color, shape, size, or weight. In general, the properties of an object are determined by the particles of the substance. The shape depends on \_\_\_\_\_.

Many of the physical properties of materials that make up objects can be measured. These properties include density, hardness, magnetism, and the ability to conduct heat and electricity.

Some physical properties depend on how much of a material you have. In other words, a large amount of a material may have different properties than a smaller amount of the material. Other physical properties do not depend on how much of the material you have. They show that no matter how large or small something is, it has the same physical property or always the same \_\_\_\_\_.

**LEARNING OBJECTIVE**  
Identify the physical and chemical properties of substances and distinguish between physical and chemical changes.

**LEARNING CHECK**  
1. Identify the three physical properties of matter.

\_\_\_\_\_

\_\_\_\_\_

\_\_\_\_\_

**LEARNING CHECK**  
2. Identify the chemical properties of matter.

\_\_\_\_\_

\_\_\_\_\_

\_\_\_\_\_

# Chapter 2 Properties Matter Wordwise Answers

**Ensheng Dong**



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