

- 4 Chlorine has two isotopes, ${}^{35}_{17}\text{Cl}$ and ${}^{37}_{17}\text{Cl}$.
- (a) What do you understand by the term *isotope*? [1]
- (b) Write down the number of neutrons in each of these isotopes of chlorine. [2]
- (c) Draw and label a diagram of an atom of chlorine to show:
- the nucleus
 - the electron shells, with the correct number of electrons in each shell. [3]
- (d) A chloride ion has eight electrons in its outer shell. What is the importance of this electronic configuration? [1]

(Paper 3)

Chapter 2 Section Review Questions

Kaplan Test Prep



Chapter 2 Section Review Questions:

Unveiling the Magic of Words: A Overview of "**Chapter 2 Section Review Questions**"

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