

Name: _____

Date: _____

Chapter 6 Study Guide: Periodicity

1. Describe the historical development of the periodic table (Newland, Meyer, Mendeleev, Mosely) Describe the contributions made by each of the scientist.

Newland – First to order the elements in a periodic table –developed the law of octaves

Meyers – Ordered elements by atomic mass but is given little credit for the current periodic table

Mendeleev – ordered elements by atomic mass and is given the most credit for the current periodic table because he left holes in his table for undiscovered elements and described their properties

Mosely – ordered elements by atomic number (modern periodic table)

2. Fill in the blanks below:

The elements in the periodic table are arranged in order of increasing Atomic

Number. The vertical columns are called Groups or

Families. The horizontal rows are called Periods. The

elements within the same Family have the same chemical properties due to the number of their Valence Electrons. Metals are located on the

Left side of the periodic table. The nonmetals are located on the

Right side of the periodic table. The metalloids are located on

the Stair step between the metals and nonmetals.

3. Describe the general properties of the following groups of elements on the periodic table:

- a. Alkali Metals

1 valence electrons, metals, highly reactive, s-block

- b. Alkaline Earth metals

2 valence electrons, metals, reactive, s-block

- c. Halogens

7 valence electrons, high reactive, nonmetals, p-block

- d. Noble Gases

8 noble electrons, nonreactive, nonmetals, all gases, p-block

- e. Transition Metals

Variable valence electrons, metals, malleable, ductile, metals, conductive

- f. Inner Transition Metals

Variable valence electrons, metals, f-block

Chapter 6 Periodic Table Answers

Thomas Evangelist



Chapter 6 Periodic Table Answers:

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Materials Principles and Practice Charles Newey, Graham Weaver, 2013-10-22 Materials Principles and Practice deals with materials science in the technological context of making and using materials Topics covered include the nature of materials such as crystals an atomic view of solids temperature effects on materials and the mechanical and chemical properties of materials This book is comprised of seven chapters and begins with an overview of the properties of different kinds of material the ways in which materials can be shaped and the uses to which they can be put The next chapter

describes the state of matter as a balance between the tendencies of atoms to stick together by chemical bonding or rattle apart by thermal agitation paying particular attention to ionic bonds and ionic crystals the structure and properties of polymers and transition metals The reader is also introduced to how the structure of materials especially microstructure can be manipulated to give desired properties via thermal mechanical and chemical agents of change This text concludes by describing the chemistry of processing and service of various materials Exercises and self assessment questions with answers are given at the end of each chapter together with a set of objectives This monograph will be a valuable resource for students of materials science and the physical sciences *New York State Regents Exam* Kaplan, Kaplan Educational Center Staff, 2000-02 Kaplan's guides to the New York State Regents Exams come complete with a comprehensive review of all the tested material plus Kaplan's exclusive test taking strategies This powerful combination makes the New York State Regents Exam Chemistry Second Edition a highly effective way for you to score higher on this very challenging test Are you ready for the New York State Regents Chemistry exam You will be with Kaplan's proven plan for success STEP 1 Take a Diagnostic Test The results of this test will outline your strengths and weaknesses You will find out the exact areas on which you need to focus your preparation STEP 2 Review the Tested Material Kaplan takes you through each section step by step providing you with effective tips and strategies to successfully answer every type of question STEP 3 Practice with Real Regents Exams Practice makes perfect And with this book you'll practice with the real thing actual Regents exams including detailed explanations for every answer and an analysis of your performance STEP 4 Succeed on the Test Follow Kaplan's plan for success on the Regents and you will score higher In fact we guarantee it See details inside [Green Materials and Environmental Chemistry](#) Abu Zahrim Yaser, Poonam Khullar, A. K. Haghi, 2021-03-24 The world faces significant challenges as the population and consumption continue to grow while nonrenewable fossil fuels and other raw materials are depleted at ever increasing rates Moreover environmental consciousness and a penchant for thinking in terms of material cycles have caught on with consumers the use of environmentally compatible materials and production methods is desired This volume *Green Materials and Environmental Chemistry New Production Technologies Unique Properties and Applications* takes a technical approach to address these issues using green design and analysis This book provides an overview of the latest developments in environmental chemistry and sustainable materials written by experts in their respective research areas This interdisciplinary volume offers research with the aim to minimize environmental impacts across all lifecycle phases in the design and engineering of products processes and systems as just one possible approach to addressing the larger issue of sustainability that includes environmental economic and social aspects [Science in Your World: Teacher resource masters](#) Jay K. Hackett, 1991 *McGraw-Hill's SAT Subject Test Chemistry, 3rd Edition* Thomas Evangelist, 2011-02-03 Expert guidance on the Chemistry exam Many colleges and universities require you to take one or more SAT II Subject Tests to demonstrate your mastery of specific high school subjects McGraw Hill's SAT Subject Test Chemistry is written by experts in

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