

CHAPTER 2

Force and Weight:

$$F = ma / g_c, \quad W = mg / g_c$$

Standard Acceleration of Gravity:

$$g = 9.80665 \text{ m/s}^2 = 980.665 \text{ cm/s}^2 = 32.174 \text{ ft/s}^2$$

$$g_c \text{ (gravitational conversion factor)} = 1 \text{ kg}\cdot\text{m}\cdot\text{s}^{-2}/\text{N} = 32.174 \text{ lb}_m\cdot\text{ft}\cdot\text{s}^{-2}/\text{lb}_f = 980.665 \text{ g}_m\cdot\text{cm}\cdot\text{s}^{-2}/g_f$$

Two Point Linear Interpolation:

$$y = y_1 + \frac{x - x_1}{x_2 - x_1} (y_2 - y_1)$$

Finding a Straight Line:

$$y = ax + b, \quad \text{SLOPE } a = \frac{y_2 - y_1}{x_2 - x_1}, \quad \text{INTERCEPT } b = \frac{y_1 x_2 - x_1 y_2}{x_2 - x_1}$$

Method of Least Squares:

$$a = \frac{\sum xy}{\sum x^2}, \quad b = \frac{\sum y}{\sum x}$$

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$$\text{SLOPE } a = \frac{y_2 - y_1}{x_2 - x_1}, \quad \text{INTERCEPT } b = \frac{y_1 x_2 - x_1 y_2}{x_2 - x_1}$$

CHAPTER 3

SI = g/cm³

$$\rho_{\text{water}} = 1.000 \text{ g/cm}^3 = 1000 \text{ kg/m}^3 = 62.43 \text{ lb}_m/\text{ft}^3$$

Mass Fraction $x_i = \frac{\text{mass of } A_i}{\text{total mass}}, \quad \text{Mole Fraction } y_i = \frac{\text{mole of } A_i}{\text{total mole}}$

Average Molecular Weight $\bar{M} = \frac{\sum n_i M_i}{\sum n_i} = \frac{\text{total mass of sample}}{\text{total moles in sample}}$

$$\bar{M} = x_1 M_1 + x_2 M_2 + \dots = \frac{\sum x_i M_i}{\sum x_i}$$

$$\frac{1}{\bar{M}} = \frac{x_1}{M_1} + \frac{x_2}{M_2} + \dots = \frac{\sum \frac{x_i}{M_i}}{\sum x_i}$$

Pressure:

Hydrostatic Pressure: $P = \rho g h$

Head of Fluid: $P(\text{head of fluid}) = \rho g h_{\text{fluid}}$

$P_{\text{total}} = P_{\text{atm}} + P_{\text{static}}$

General Manometer Equation: $P_1 + \rho_1 \frac{P_1}{g} z_1 = P_2 + \rho_1 \frac{P_2}{g} z_2 + \rho_2 \frac{P_2}{g} z_2$

Differential Manometer Equation: $P_1 - P_2 = (\rho_2 - \rho_1) \frac{P_2}{g} z_2$

Manometer Formula for Gases: $P_1 - P_2 = 0$

Temperature:

$32^\circ\text{F} = 0^\circ\text{C} = 273.15 \text{ K}, \quad 212^\circ\text{F} = 100^\circ\text{C} = 373.15 \text{ K}, \quad 0^\circ\text{F} = -17.78^\circ\text{C} = 255.37 \text{ K}, \quad 100^\circ\text{F} = 37.78^\circ\text{C} = 310.93 \text{ K}$

Chemical Engineering Formula Sheet

David Baud



Chemical Engineering Formula Sheet:

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Table of Contents Chemical Engineering Formula Sheet

1. Understanding the eBook Chemical Engineering Formula Sheet
 - The Rise of Digital Reading Chemical Engineering Formula Sheet
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemical Engineering Formula Sheet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemical Engineering Formula Sheet
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemical Engineering Formula Sheet
 - Personalized Recommendations
 - Chemical Engineering Formula Sheet User Reviews and Ratings
 - Chemical Engineering Formula Sheet and Bestseller Lists

5. Accessing Chemical Engineering Formula Sheet Free and Paid eBooks
 - Chemical Engineering Formula Sheet Public Domain eBooks
 - Chemical Engineering Formula Sheet eBook Subscription Services
 - Chemical Engineering Formula Sheet Budget-Friendly Options
6. Navigating Chemical Engineering Formula Sheet eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemical Engineering Formula Sheet Compatibility with Devices
 - Chemical Engineering Formula Sheet Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemical Engineering Formula Sheet
 - Highlighting and Note-Taking Chemical Engineering Formula Sheet
 - Interactive Elements Chemical Engineering Formula Sheet
8. Staying Engaged with Chemical Engineering Formula Sheet
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemical Engineering Formula Sheet
9. Balancing eBooks and Physical Books Chemical Engineering Formula Sheet
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemical Engineering Formula Sheet
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemical Engineering Formula Sheet
 - Setting Reading Goals Chemical Engineering Formula Sheet
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemical Engineering Formula Sheet
 - Fact-Checking eBook Content of Chemical Engineering Formula Sheet
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

-
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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