

# SALT ANALYSIS

## 1) For $MgSO_4$ - magnesium Sulfate :-

**Aim:** To analyze the given salt for an anion and one cation present in it.

**Material Required:** Boiling Tube, Test Tubes, Test Tube Holder, Test Tube Stand, delivery tube, Corks, Filter paper, etc.

### Procedure:-

S.No.	Test	Observation	Inference
1)	Colour	White	
2)	Odour		
3)	Flame test	Greenish white	
4)	Test $H_2SO_4$ Sol (Salt + conc $H_2SO_4$ )	No ppt. form	
5)	Test $H_2SO_4$ Sol (Salt + conc $H_2SO_4$ )	No.	

## 2) For $Fe^{2+}Cl^-$ , Ferrous chloride :-

**Aim:** To analyze the given salt for an anion and one cation present in it.

**Material Required:** Boiling Tube, Test Tube, Test Tube Holder, Test Tube Stand, delivery tube, Corks, Filter paper, etc.

### Procedure:-

S.No.	Test	Observation	Inference
		Brown reddish	$Fe^{2+}$ may be present
		Colourless	$NO_3^-$ , $S^{2-}$ , $CO_3^{2-}$ are absent
			$Ca^{2+}$ , $S^{2-}$ , $SO_4^{2-}$ , $NO_3^-$ are absent
			$Cl^-$ may be present
			$Cl^-$ confirmed

Class 12th

Chemistry Practical File

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**Dr. R.D. Madan & B S Bisht**



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## **Table of Contents Chemistry Practical File Class 12 Salt Analysis**

1. Understanding the eBook Chemistry Practical File Class 12 Salt Analysis
  - The Rise of Digital Reading Chemistry Practical File Class 12 Salt Analysis
  - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Practical File Class 12 Salt Analysis
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemistry Practical File Class 12 Salt Analysis
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Practical File Class 12 Salt Analysis
  - Personalized Recommendations
  - Chemistry Practical File Class 12 Salt Analysis User Reviews and Ratings
  - Chemistry Practical File Class 12 Salt Analysis and Bestseller Lists

5. Accessing Chemistry Practical File Class 12 Salt Analysis Free and Paid eBooks
  - Chemistry Practical File Class 12 Salt Analysis Public Domain eBooks
  - Chemistry Practical File Class 12 Salt Analysis eBook Subscription Services
  - Chemistry Practical File Class 12 Salt Analysis Budget-Friendly Options
6. Navigating Chemistry Practical File Class 12 Salt Analysis eBook Formats
  - ePub, PDF, MOBI, and More
  - Chemistry Practical File Class 12 Salt Analysis Compatibility with Devices
  - Chemistry Practical File Class 12 Salt Analysis Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chemistry Practical File Class 12 Salt Analysis
  - Highlighting and Note-Taking Chemistry Practical File Class 12 Salt Analysis
  - Interactive Elements Chemistry Practical File Class 12 Salt Analysis
8. Staying Engaged with Chemistry Practical File Class 12 Salt Analysis
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chemistry Practical File Class 12 Salt Analysis
9. Balancing eBooks and Physical Books Chemistry Practical File Class 12 Salt Analysis
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chemistry Practical File Class 12 Salt Analysis
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Practical File Class 12 Salt Analysis
  - Setting Reading Goals Chemistry Practical File Class 12 Salt Analysis
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Practical File Class 12 Salt Analysis
  - Fact-Checking eBook Content of Chemistry Practical File Class 12 Salt Analysis
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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