

**Chemistry 12  
Homework Assignment 3-1**

46

Name KEY

Due Date \_\_\_\_\_

**Correct and Hand In By** \_\_\_\_\_

*This worksheet covers material from Tutorials 7 and 8, class notes and SW pages 73-87.*

**These questions will be marked and the mark recorded in the "Homework" section in the mark book. One set of corrections will be allowed. Get help if you need it.**

1. Decide whether each of the following are electrolytes or not (yes or no) when dissolved in water. (5 marks)

5

- a)  $\text{AgNO}_3$  ..... yes
- b)  $\text{CH}_3\text{CH}_2\text{OH}$  ..... no
- c)  $\text{NH}_4\text{Cl}$  ..... yes
- d)  $\text{CH}_3\text{CH}_2\text{COOH}$  ..... yes
- e)  $\text{Ba}(\text{OH})_2$  ..... yes

2. Give three examples of ionic compounds (formulas must be correct) (3 marks)

3

- 1. \_\_\_\_\_
- 2. \_\_\_\_\_
- 3. \_\_\_\_\_

3. Give two examples of molecular compounds (formulas must be correct) (2 marks)

2

- 1. \_\_\_\_\_
- 2. \_\_\_\_\_

Define a saturated solution (1 mark) \_\_\_\_\_

1

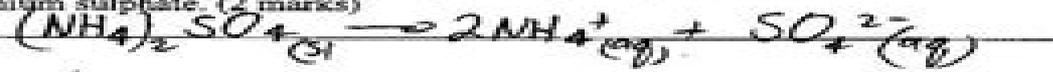
Solubility is the equilibrium concentration of a substance in solution at a given temperature. (1 mark)

1

6. Write a net ionic equation (balanced, all charges, subscripts, arrow(s) correct, correct order etc.) for dissolving ammonium sulphate. (2 marks)

2

14  
14



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**The Princeton Review**



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### **Table of Contents Chemistry Unit 3 Test Answer Key**

1. Understanding the eBook Chemistry Unit 3 Test Answer Key
  - The Rise of Digital Reading Chemistry Unit 3 Test Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Unit 3 Test Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemistry Unit 3 Test Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Unit 3 Test Answer Key
  - Personalized Recommendations
  - Chemistry Unit 3 Test Answer Key User Reviews and Ratings
  - Chemistry Unit 3 Test Answer Key and Bestseller Lists
5. Accessing Chemistry Unit 3 Test Answer Key Free and Paid eBooks
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- Chemistry Unit 3 Test Answer Key Budget-Friendly Options
- 6. Navigating Chemistry Unit 3 Test Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Chemistry Unit 3 Test Answer Key Compatibility with Devices
  - Chemistry Unit 3 Test Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chemistry Unit 3 Test Answer Key
  - Highlighting and Note-Taking Chemistry Unit 3 Test Answer Key
  - Interactive Elements Chemistry Unit 3 Test Answer Key
- 8. Staying Engaged with Chemistry Unit 3 Test Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chemistry Unit 3 Test Answer Key
- 9. Balancing eBooks and Physical Books Chemistry Unit 3 Test Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chemistry Unit 3 Test Answer Key
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Unit 3 Test Answer Key
  - Setting Reading Goals Chemistry Unit 3 Test Answer Key
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Unit 3 Test Answer Key
  - Fact-Checking eBook Content of Chemistry Unit 3 Test Answer Key
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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