Chapter 3 Chemistry Test: The Atom

Multiple Choice
Identify the choice that best completes the statement or answers the question.

_____ 1. The law of conservation of mass follows from the concept that
a. atoms are indivisible.
b. atoms of different elements have different properties.
c. matter is composed of atoms.
d. atoms can be destroyed in chemical reactions.

_____ 2. Who explained the behavior of positively charged particles being deflected from a metal foil as the nucleus?
a. Ernest Rutherford
b. John Dalton
c. James Chadwick
d. Niels Bohr

_____ 3. In the gold foil experiment, most of the particles fired at the foil
a. bounced back.
b. passed through the foil.
c. were absorbed by the foil.
d. combined with the foil.

_____ 4. What did Rutherford conclude about the structure of the atom?
a. An atom is indivisible.
b. Electrons make up the center of an atom.
c. An atom carries a positive charge.
d. An atom contains a small, dense, positively charged central region.

_____ 5. A nuclear particle that has about the same mass as a proton, but with no electrical charge, is called a(n)
a. nucleus.
b. neutron.
c. electron.
d. isotope.

_____ 6. Which part of an atom has a mass approximately equal to $\frac{1}{1836}$ of the mass of a common hydrogen atom?
a. nucleus
b. electron
c. proton
d. electron cloud

_____ 7. The mass of a neutron is
a. about the same as that of a proton.
b. about the same as that of an electron.
c. double that of a proton.
d. double that of an electron.

_____ 8. The nucleus of most atoms is composed of
a. tightly packed protons.
b. tightly packed neutrons.
c. tightly packed protons and neutrons.
d. loosely concentrated protons and electrons.

_____ 9. An aluminum isotope consists of 13 protons, 13 electrons, and 14 neutrons. Its mass number is
a. 13.
b. 14.
c. 27.
d. 40.

_____ 10. Isotopes are atoms of the same element that have different
a. principal chemical properties.
b. masses.
c. numbers of protons.
d. numbers of electrons.

_____ 11. Atoms of the same element that have different masses are called
a. moles.
b. isotopes.
c. nuclides.
d. neutrons.

_____ 12. Most of the volume of an atom is occupied by the
a. nucleus.
b. nuclides.
c. electron cloud.
d. protons.

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Modern Chemistry Chapter 3 Test Answer Key

Sam S. Blanc

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