
**IFMA FMP PROJECT MANAGEMENT EXAM QUESTIONS
AND ANSWERS VERIFIED ANSWERS WITH RATIONALES |
UPDATED 2024 | GRADED A |**

If an organization has narrowed the list of contractors submitting proposals but is still having trouble deciding between them, what could help? - **CORRECT ANSWER** The correct answer is: Bidder presentations

Which statement is true of the Execute, Monitor and Control phases of the project model? - **CORRECT ANSWER** The correct answer is: These phases are often concurrently or iteratively performed.

When project managers determine the level of data to collect for monitoring and controlling project work, what do they not need to verify? - **CORRECT ANSWER** The correct answer is: Project work delays due to data collection are scheduled.

What do project managers submit when they suggest revising the project management plan to make it feasible given the current situation? - **CORRECT ANSWER** The correct answer is: change order.

What is developed to review a final deliverable from an entire system perspective? - **CORRECT ANSWER** The correct answer is: Checklist reviews

Which statement is true about the documentation for project closure? - **CORRECT ANSWER** The correct answer is: Documentation may include operations and maintenance manuals.

A project manager wants to minimize risk and provide some independent checks and balances on contractor work. Which contract model should the organization use? - **CORRECT ANSWER** Construction management Your answer is correct.

Construction management is the best choice because it uses an independent consultant to supervise separate contracts with design firms and construction firms, which allows for some independent review. (Reference: 3-2-10)

According to the critical path method (CPM) of scheduling, what is the critical path for any project? - **CORRECT ANSWER** the longest set of sequential tasks in the project. Your answer is correct.

The critical path is the longest set of sequential tasks in a project and indicates the

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Ifma Fmp Exam Questions: Bestsellers in 2023 The year 2023 has witnessed a remarkable surge in literary brilliance, with numerous compelling novels enthralling the hearts of readers worldwide. Lets delve into the realm of top-selling books, exploring the captivating narratives that have captivated audiences this year. Ifma Fmp Exam Questions : Colleen Hoover's "It Ends with Us" This poignant tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover masterfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can triumph. Ifma Fmp Exam Questions : Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This intriguing historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids captivating storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Discover the Magic : Delia Owens "Where the Crawdads Sing" This mesmerizing coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens spins a tale of resilience, survival, and the transformative power of nature, captivating readers with its evocative prose and mesmerizing setting. These top-selling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of engaging stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a brilliant and thrilling novel that will keep you guessing until the very end. The novel is a warning tale about the dangers of obsession and the power of evil.

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web google classroom for the following redox equation $\text{MnO}_4 + 4\text{A} + \text{AQ} + \text{SO}_2 + 2\text{G} + \text{Mn} + 2\text{AQ} + \text{HSO}_4 + 4\text{A} + \text{AQ}$ what will be the stoichiometric coefficient of HSO_4^- if the above equation is balanced in an acidic medium stuck use a hint report a problem do 4 problems

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web solution 1 the two half reactions balanced as if in acidic solution $2\text{CN}^- + \text{Au} \rightarrow 2\text{C}_2\text{O}_4^{2-} + \text{Au}^{3+}$ make electrons equal convert to basic solution $4\text{CN}^- + 2\text{Au} \rightarrow 2\text{C}_2\text{O}_4^{2-} + 2\text{Au}^{3+}$ multiplied by a factor of 2 $2\text{C}_2\text{O}_4^{2-} + 2\text{H}_2\text{O} \rightarrow \text{H}_2\text{O}_2 + 2\text{OH}^-$ the final answer

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web we can balance it for charge by adding two electrons to the right side of the equation so that the net charge on each side is 0 oxidation $\text{Ni} \rightarrow \text{Ni}^{2+} + 2\text{e}^-$ now that the oxidation half reaction is balanced it tells us that two electrons are produced for every atom of nickel oxidized but where do those electrons go

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web aug 29 2023 some points to remember when balancing redox reactions the equation is separated into two half equations one for oxidation and one for reduction the equation is balanced by adjusting coefficients and adding H_2O H^+ and e^- in this order balance the atoms in the equation apart from o and h

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web in the redox equation below that occurs in our stomach which of the following pairs identifies the reducing and oxidizing agents respectively $\text{C}_6\text{H}_8\text{O}_6 + 2\text{H}_2\text{NO}_2 \rightarrow \text{C}_6\text{H}_6\text{O}_6 + 2\text{H}_2\text{O} + 2\text{NO}$ choose 1 answer $\text{C}_6\text{H}_8\text{O}_6$ and NO_2 a $\text{C}_6\text{H}_8\text{O}_6$ and NO_2 no

2 and c 6 h 8 o 6 b no 2 and c 6 h 8 o 6 h and no 2 c h and

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