

Chemistry Chapter 8 Exam

Multiple Choice

1. Which of the following is not a Lewis acid?

- _____ a. CH_3COOH
- _____ b. Lewis Acids
- _____ c. Carbonyl Compounds
- _____ d. Alcohols
- _____ e. Metal Ions
- _____ f. Simple Molecules
- _____ g. Molecule Molecules
- _____ h. Lewis Acids
- _____ i. Hydrogen Ions
- _____ j. None of the above

- _____ k. bonded to electronegative atoms that are more electronegative than the central atom
- _____ l. single group of molecules bonded to one
- _____ m. the energy released when one mole of an ionic crystalline compound is dissolved in water
- _____ n. molecules in water
- _____ o. coordination bonds are part of the structure of a complex ion
- _____ p. the energy released in a bond is always less than the energy required to break it
- _____ q. bonded group of atoms that are held together by covalent bonds
- _____ r. results from the sharing of electron pairs between atoms
- _____ s. chemical bonding that results from the electrical attraction between positive and negative ions
- _____ t. mutual electrical attraction between positive and negative ions of opposite charge, producing the ionic compound

Multiple Choice

2. The charge on the ion is _____

- _____ a. always positive
- _____ b. always negative
- _____ c. either positive or negative
- _____ d. zero

3. In many compounds, atoms of more electronegative elements form bonds with the atoms of less electronegative elements. The bond between _____

- _____ a. is _____
- _____ b. is _____
- _____ c. is _____
- _____ d. is _____

4. Which of the following is not a Lewis acid? _____

- _____ a. CH_3COOH
- _____ b. Lewis Acids
- _____ c. Carbonyl Compounds
- _____ d. Alcohols
- _____ e. Metal Ions
- _____ f. Simple Molecules
- _____ g. Molecule Molecules
- _____ h. Lewis Acids
- _____ i. Hydrogen Ions
- _____ j. None of the above

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