

Name: .....

### ATOMIC STRUCTURE

1 Use the words in the box to answer the following questions.

**electron, neutron, nucleus, proton, shells**



- (a) The central part of an atom is called the .....
- (b) The ..... has a negative charge
- (c) The ..... has no charge
- (d) The nucleus contains the ..... and the .....
- (e) Electrons are found in .....
- (f) The ..... and the ..... have the same mass.

2 An atom contains 32 protons and 48 neutrons.

- (a) What is its atomic number? .....
- (b) How many electrons does it have? .....
- (c) What is the mass of the atom? .....

3 Complete the table to show the missing numbers of particles in the atoms A - E.

Atom	Protons	Neutrons	Electrons	Mass
A	20			40
B			21	45
C		24		44
D	19	21		
E			22	45

Which two atoms are isotopes of the same element? ..... and .....

4 Complete the table to show the missing information:

Atom	Atomic number	Electron arrangement	Group in the Periodic Table	Period in the Periodic Table
P		2, 8, 7		
Q	11			
R			4	2
S			3	3

# Atomic Structure Test Questions

**Peter J. Mikulecky, Michelle Rose  
Gilman, Kate Brutlag**



## Atomic Structure Test Questions:

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