

9/2/16

S. Sai Krishna

Date \_\_\_\_\_  
Page \_\_\_\_\_  
ClassmateThermochemistry

$$Q = \Delta E - W$$

$$Q = \Delta E - (-P\Delta V)$$

$$Q = \Delta E + P\Delta V$$

- At constant Volume.

$$Q = \Delta E$$

- At const. Pressure.

$$Q_p = \Delta E + P\Delta V$$

$$H = \Delta E + P\Delta V$$

$$H = \Delta E + \Delta nRT$$

$\Delta n =$  no. of moles of Prod  
- no. of moles of react.

→ For  $\Delta H$ .

- If  $\Delta H$  is -ve. reaction is exothermic.
- If  $\Delta H$  is +ve. reaction is endothermic.

Q. The heat of combustion of ethane at 17°C and const volume is  $-322.1 \text{ kJ}$ . Calculate the heat of combustion at constant pressure.



# Be Engineering Chemistry Notes Pdf 2016

**JE Gale**



**Be Engineering Chemistry Notes Pdf 2016:**

## Whispering the Strategies of Language: An Emotional Journey through **Be Engineering Chemistry Notes Pdf 2016**

In a digitally-driven world where screens reign supreme and instant connection drowns out the subtleties of language, the profound strategies and emotional nuances hidden within words usually move unheard. However, situated within the pages of **Be Engineering Chemistry Notes Pdf 2016** a interesting fictional value pulsating with fresh thoughts, lies an exceptional journey waiting to be undertaken. Penned by a talented wordsmith, that wonderful opus invites readers on an introspective trip, gently unraveling the veiled truths and profound influence resonating within the very cloth of every word. Within the emotional depths of this poignant review, we can embark upon a genuine exploration of the book is core subjects, dissect their captivating publishing design, and yield to the effective resonance it evokes strong within the recesses of readers hearts.

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