

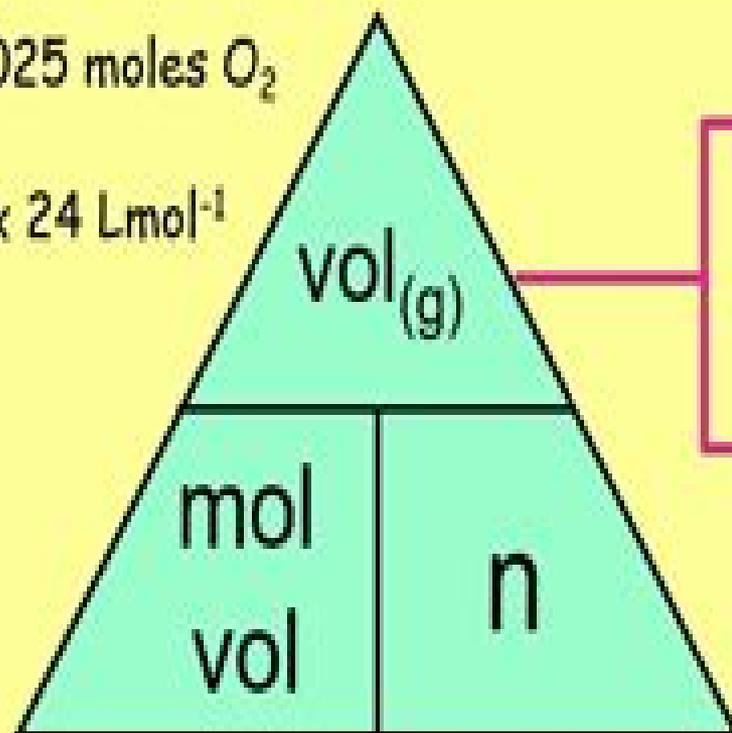
Assume molar volume is  $24\text{Lmol}^{-1}$

Calculate volume of 0.025 moles  $\text{O}_2$

$$\text{Vol} = 0.025 \text{ mol} \times 24 \text{ Lmol}^{-1}$$

$$= 0.6 \text{ L}$$

$$= 600 \text{ cm}^3$$



Gas Volume

$$\text{Vol} = n \times \text{mol.vol}$$

# Calculations Using Standard Molar Volume

**Robert J. Silbey, Robert A.  
Alberty, George A.  
Papadantonakis, Mounqi G. Bawendi**

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