

UNIT 10: PERIODIC TABLE **PERIODIC TABLE** **PERIODIC TABLE** **PERIODIC TABLE** **PERIODIC TABLE**

Section 10.1 *(continued)*

In your notebook, read aloud the modern periodic table.

Copy the letterheads in the boxes on the left below from the periodic table to complete the table on the right.

atomic mass	_____
atomic number	_____
electron configuration	_____
chemical name	_____
chemical symbol	_____

Put each item in Column A, under the letter of the matching item in Column B.

Column A

Column B

- | | |
|---|----------------------------------|
| _____ 18. A column in the periodic table | _____ a. metals |
| _____ 19. A row in the periodic table | _____ b. gases |
| _____ 20. Group 1 elements | _____ c. period |
| _____ 21. Elements that are shiny and conduct electricity | _____ d. representative elements |
| _____ 22. Group 8 elements | _____ e. transition elements |

In the space on the left, write each of the statements in terms of the statement in bold. Change the underlined word or phrase to make it true.

- _____ 23. There are two representative elements of platinum.
- _____ 24. Most nonmetals form diatomic molecules in the periodic table (or naturally).
- _____ 25. Group 1A elements except for hydrogen are located on the left side.
- _____ 26. Group 2A elements are the alkaline earth metals.
- _____ 27. Group 7A elements are highly reactive nonmetals known as halogens.
- _____ 28. Group 8A elements are inert nonmetallic elements known as noble gases.
- _____ 29. Alkaloids have properties of both metals and more reactive metals.

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A Loxley



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Table of Contents Chemistry Chapter 6 Study Guide For Content Mastery Answers

1. Understanding the eBook Chemistry Chapter 6 Study Guide For Content Mastery Answers
 - The Rise of Digital Reading Chemistry Chapter 6 Study Guide For Content Mastery Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Chapter 6 Study Guide For Content Mastery Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Chapter 6 Study Guide For Content Mastery Answers
 - User-Friendly Interface

4. Exploring eBook Recommendations from Chemistry Chapter 6 Study Guide For Content Mastery Answers
 - Personalized Recommendations
 - Chemistry Chapter 6 Study Guide For Content Mastery Answers User Reviews and Ratings
 - Chemistry Chapter 6 Study Guide For Content Mastery Answers and Bestseller Lists
5. Accessing Chemistry Chapter 6 Study Guide For Content Mastery Answers Free and Paid eBooks
 - Chemistry Chapter 6 Study Guide For Content Mastery Answers Public Domain eBooks
 - Chemistry Chapter 6 Study Guide For Content Mastery Answers eBook Subscription Services
 - Chemistry Chapter 6 Study Guide For Content Mastery Answers Budget-Friendly Options
6. Navigating Chemistry Chapter 6 Study Guide For Content Mastery Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Chapter 6 Study Guide For Content Mastery Answers Compatibility with Devices
 - Chemistry Chapter 6 Study Guide For Content Mastery Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Chapter 6 Study Guide For Content Mastery Answers
 - Highlighting and Note-Taking Chemistry Chapter 6 Study Guide For Content Mastery Answers
 - Interactive Elements Chemistry Chapter 6 Study Guide For Content Mastery Answers
8. Staying Engaged with Chemistry Chapter 6 Study Guide For Content Mastery Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Chapter 6 Study Guide For Content Mastery Answers
9. Balancing eBooks and Physical Books Chemistry Chapter 6 Study Guide For Content Mastery Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Chapter 6 Study Guide For Content Mastery Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Chapter 6 Study Guide For Content Mastery Answers
 - Setting Reading Goals Chemistry Chapter 6 Study Guide For Content Mastery Answers
 - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Chemistry Chapter 6 Study Guide For Content Mastery Answers
 - Fact-Checking eBook Content of Chemistry Chapter 6 Study Guide For Content Mastery Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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