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2. Installation

2.1. Overview

This chapter documents the procedures for installing Virtual Reactor and its license server, the Reprise License Manager (RLM). The following steps lead you through the process of getting started with Virtual Reactor:

1. [Create a CPFD Support site account](#) in order to access downloads
2. Acquire hardware that meets recommended [System Requirements](#)
3. [Plan the Client-Server Configuration](#)
4. [Download Virtual Reactor](#)
5. [Install the RLM Server](#)
6. [Install the RLM License](#)
7. [Install Virtual Reactor](#)
8. [Point Client Machines to the RLM Server](#)
9. [Install NVIDIA Drivers](#)

In order to use Virtual Reactor, you must have a valid software license installed. Virtual Reactor uses a client-server model for licensing, meaning that your license(s) can be placed on one machine, which acts as a license server, and other machines on the same network can point to and request licenses from that server. The server may also use the license(s) that it hosts.

The licensing policy for Virtual Reactor is as follows:

- Each solver license allows the user to run one instance of the Virtual Reactor solver. Optionally, add-ons to each solver license may be purchased to include Chemistry capabilities and/or Parallel capabilities.
- The number of Virtual Reactor solvers that can be run simultaneously is limited to the number of solver licenses that have been purchased.
- With any license, an unlimited number of Virtual Reactor Graphical User Interfaces (GUIs) can be open at the same time. This allows users to set up simulations, review previously set up models, and use all functions of the GUI.
- [Running a single time step](#) does not count against the number of Virtual Reactor solver licenses in use. This allows users to review a project setup at any time.
- An unlimited number of GMV post-processing instances can be open at the same time.
- If Parallel add-on licenses have been purchased, Virtual Reactor can utilize NVIDIA GPU cards to increase calculation speed. The number of simulations that can use GPU acceleration simultaneously is limited to the number of Parallel add-on licenses purchased.
- If Chemistry add-on licenses have been purchased, Virtual Reactor can include user-specified chemical reactions in the simulation. The number of simulations that can solve chemistry simultaneously is limited to the number of chemistry add-on licenses that have been purchased.
- If Parallel and Chemistry add-on licenses have been purchased, Virtual Reactor can use multiple CPU cores for parallelization of volume-average chemistry to increase calculation speed. The number of simulations that can use CPU parallel acceleration simultaneously is limited to the number of Parallel add-on licenses purchased.

2.2. System Requirements

Virtual Reactor can be used on either Linux or Windows. The following table lists minimum and recommended system requirements.

| | Minimum | Recommended |
|------------------|---|---|
| Operating System | Recent 64-bit Linux 64-bit Windows 7, 8, or 10 | 64-bit CentOS 6 (RHEL 6) or higher Windows 7 Professional 64-bit |

Barracuda User Guide

Karin Nielsen-Saines



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web by utilizing the half equivalence point on the titration curve the pka of the acid analyte was determined to be 3.655 furthermore using the stoichiometric ratio between the naoh and the acid analyte 1:1 we could experimentally determine the number of moles of the acid in the solution to be 0.00731 moles making the experimental molar mass

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 $\text{HCl(aq)} + \text{NaOH(aq)} \rightarrow \text{NaCl(aq)} + \text{H}_2\text{O(l)}$
one mole of HCl would be fully neutralized by one mole of NaOH

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web conclusion the identification of an unknown solid acid the purpose of this experiment was to determine the standardization of sodium hydroxide NaOH and oxalic acid $\text{H}_2\text{C}_2\text{O}_4 \cdot 2\text{H}_2\text{O}$ and the molar mass of an unknown solid acid using the molar mass of the standardized sodium hydroxide solution

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web specifically an acid base titration can be used to figure out the concentration of an acid or base whether an unknown acid or base is strong or weak pka of an unknown acid or pkb of the unknown base key terms titration a process where a solution of known strength is added to a certain volume of a treated sample containing an

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web conclusion this lab evaluated an acid base reaction with the volumetric analysis of a standard solution being naoh reaction with a 2nd substance $\text{H}_2\text{C}_2\text{O}_4$ aq 2NaOH aq $\text{Na}_2\text{C}_2\text{O}_4$ $4\text{H}_2\text{O}$ to differentiate an acid from a base in a solution a ph scale is used the ph scale ranges from 0 to 14 a ph of 7 is neutral in an acid base reaction

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web this point of titration is referred to as the equivalence point considering the 1 1 stoichiometry of this acid base reaction NaOH aq $\text{C}_6\text{H}_4\text{COOH}$ aq $\text{C}_6\text{H}_4\text{COONa}$ aq H_2O l the point of equivalence is the point of titration when the number of moles of naoh added is equal to the number of moles of khp in the

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