



# COVALENT BONDING

## Practice Problems

Do your answers to the following problems.

### SECTION 8.1 MOLECULAR COMPOUNDS

- Draw the Lewis structure for each of the following molecules.
  - $\text{H}_2\text{O}$
  - $\text{CO}_2$
  - $\text{H}_2\text{S}$
  - $\text{CH}_4$
  - $\text{NH}_3$
  - $\text{H}_2\text{O}_2$
  - $\text{C}_2\text{H}_6$
  - $\text{C}_2\text{H}_4$
  - $\text{C}_2\text{H}_2$
- Draw the Lewis structure for each of the following molecules.
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  - $\text{CO}_2$
  - $\text{H}_2\text{S}$
  - $\text{CH}_4$
  - $\text{NH}_3$
  - $\text{H}_2\text{O}_2$
  - $\text{C}_2\text{H}_6$
  - $\text{C}_2\text{H}_4$
  - $\text{C}_2\text{H}_2$
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  - $\text{C}_2\text{H}_4$
  - $\text{C}_2\text{H}_2$
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  - $\text{CH}_4$
  - $\text{NH}_3$
  - $\text{H}_2\text{O}_2$
  - $\text{C}_2\text{H}_6$
  - $\text{C}_2\text{H}_4$
  - $\text{C}_2\text{H}_2$

### SECTION 8.2 THE NATURE OF COVALENT BONDING

- Draw the Lewis structure for each of the following molecules.
 

$\text{H}-\text{F}$
- Draw the Lewis structure for each of the following molecules.
 

$\begin{array}{c} \text{H} \\ | \\ \text{H}-\text{C}-\text{H} \\ | \\ \text{H} \end{array}$
- Draw the Lewis structure for each of the following molecules.
 

$\text{H}-\text{C}\equiv\text{C}-\text{H}$
- Draw the Lewis structure for each of the following molecules.
 

$\text{H}-\text{C}\equiv\text{C}-\text{H}$
- Draw the Lewis structure for each of the following molecules.
 

$\text{H}-\text{C}\equiv\text{C}-\text{H}$



### SECTION 8.3 BONDING: THE ORBITALS

- Draw the Lewis structure for each of the following molecules.
  - $\text{H}_2\text{O}$
  - $\text{CO}_2$
  - $\text{H}_2\text{S}$
  - $\text{CH}_4$
  - $\text{NH}_3$
  - $\text{H}_2\text{O}_2$
  - $\text{C}_2\text{H}_6$
  - $\text{C}_2\text{H}_4$
  - $\text{C}_2\text{H}_2$
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  - $\text{H}_2\text{O}$
  - $\text{CO}_2$
  - $\text{H}_2\text{S}$
  - $\text{CH}_4$
  - $\text{NH}_3$
  - $\text{H}_2\text{O}_2$
  - $\text{C}_2\text{H}_6$
  - $\text{C}_2\text{H}_4$
  - $\text{C}_2\text{H}_2$

# Covalent Bonding Pearson Answers

**Richard J. D. Tilley**



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